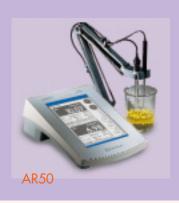
Accumet®

Electrochemistry Handbook



Accumet® Benchtop Meters pH/mV/lon/Conductivity









Advanced technology makes them simple, accurate, versatile.

Exceptional ease of use: clear language prompts, "HELP" screens and error messages guide you through the measurement process

Large, easy-to-read LCD readouts with reliable, proven touchscreen technology:

responds to a light tap, even with gloves or a pencil

Easily customizable screens: let you display only the data you want and need

Two series of meters— Basic and Research Models:

to suit your requirements

Fisher's family of Accumet Benchtops has been completely redesigned. Now, they're not only accurate and versatile, but also the easiest to use, most user-friendly pH/mV/ion/conductivity meters you can buy.

From their sleek, compact design to their oversize LCD readouts to their innovative, highly reliable touchscreen technology (Research Models only), Accumet Benchtop Meters put a wealth of information at your fingertips. Just look at what they offer you.

Standardization is ultra simple

Standardizing Accumet Benchtop Meters is so simple a child can do it. Here's all you do:

- 1. Press "STANDARDIZE" to enter the standardization mode
- 2. Press "STANDARDIZE" again to perform the standardization

That's all there is to it. In response to your commands, the meter automatically recognizes the buffer, adjusts the buffer value for temperature, waits for a stable reading, enters the data, verifies proper electrode performance, and calibrates the meter.

Throw away your user manual?

We'd certainly never advise you to do it! But Accumet Benchtop Meters are so easy to use, you probably could. Accumet Benchtop Meters feature simple, intuitive operation, on-screen operational prompts, and context-specific "HELP" screens (Accumet Research Models only) designed to guide you through the meter's operation in plain, easily understandable language.

Plus, not only are Accumet Benchtop Meters accurate, versatile, and easy to use, they're backed by a full two-year warranty. Should any Accumet Meter fail to perform to specification under normal use during the first two years, it will be repaired or replaced at no charge.

Easily customize screens and measurement modes to match your requirements

There are two ways to customize the screens and measurement modes on Accumet Benchtop Meters:

Choosing the System Setup Options menu on



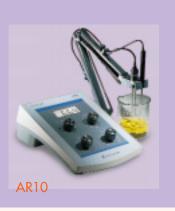
Research Model Meters lets you customize the screen display to show exactly the parameters that suit your needs. You can edit the date, screen configuration, time, beeper status, and print configuration.

Customizing within each meter's operational modes (pH/mV/ion/conductivity) lets you select appropriate operating parameters, resolution, buffer sets, and similar settings (different for each operating mode) to match your requirements.

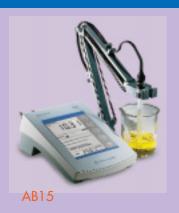
You can also choose either of two procedural levels: Basic Level displays a reduced set of information and restricts user access to a limited list of setup parameters. Advanced Level allows user-configurable output and access to all setup parameters. Using this feature enables scientists to configure the meter in Advanced Level and have a technician perform measurements in Basic Level, limiting the technician's ability to change setup parameters.

Table of Contents

Accumet Benchtop Meters	2-11	Accumet Portable Meters	12-13
pH Theory and Measurements	14-15	Conductivity Theory and Measurements	16
Metallic Electrodes, ORP Theory and Measurements	16	ISE Theory and Measurements	16-18
Accumet Electrode Specifications	19-23	Buffers, Solutions, Accessories	24







pHree Trial Offer makes it so easy to try one

We're so sure you'll agree that Accumet[®] Meters offer the best performance, versatility, and value available that we invite you to try one FREE for 30 days with no risk and no obligation. To schedule your Accumet pHree Trial, contact your Fisher Sales Representative.

Choose the meter that makes your job easiest

Two series of Accumet Benchtop Meters are available to match any laboratory's needs:

Accumet Research (AR) Benchtop Meters

- Five single- and multi-function models
- Unparalleled selection of user options lets you customize the meter to your application
- Extra-large, scratch-resistant LCD touch screens (all models except AR10) respond to a light tap, even with gloves or a pencil
- Two procedural levels to fit your needs
- Configurable display shows only data you need
- Plain language prompts and context-specific "HELP" screens guide you through
- Data acquisition via bidirectional communication with computer or printer
- Multiple language option: English, German, French, or Spanish built in
- Rugged ABS plastic housings resist impacts and chemicals

Accumet Basic (AB) Benchtop Meters

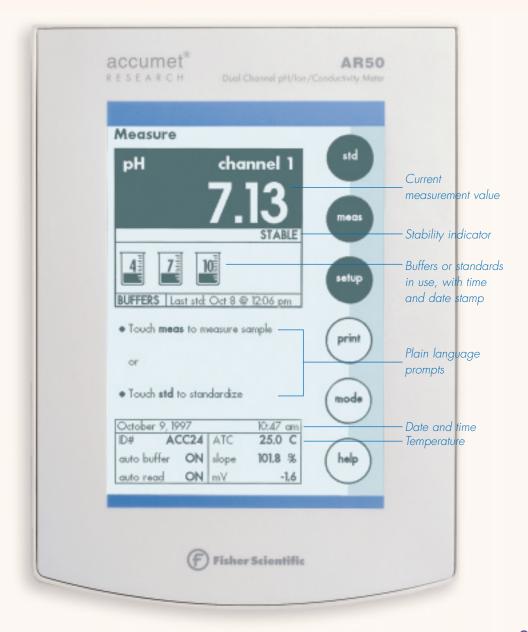
- Two models to choose from
- Accurate pH, mV, temperature (AB15) and conductivity measurement (AB30) at an affordable price
- Extra-large LCD shows results clearly
- Five soft-touch membrane switches control all operations
- On-screen plain language prompts and error messages make measurement easier

The biggest LCD readouts in their class—plus reliable, user-friendly touchscreen technology

No more squinting at tiny displays. With the biggest LCD readouts in their class, Accumet Benchtop Meters display the information you need precisely and clearly.

Accumet Basic Benchtop Meters feature a softtouch, membrane switch keypad for operation. It's tough, chemical-resistant, and simple to use. What's more, Accumet Research Meters (except Model AR10) are the only meters of their type to offer exciting, easy-to-use touchscreen technology—proven reliable in thousands of retail applications worldwide. ("TOUCH" areas of the screen will withstand millions of touches.)

Just touch the scratch-resistant screen with your fingertip to display the information you need, including current measurement value, stability indicator, buffers or standards in use with time/date stamp, plain language prompts, date, time, temperature—and much more!



The Accumet® Research Model AR50 pH/mV/lon/Conductivity Meter



Now, it's easy to put a complete electrochemical lab at your fingertips

- Powerful and versatile—make two different measurements simultaneously
- Exceptionally easy to use
- Extra-large, scratch- and chemicalresistant LCD readout with dual display
- Reliable touchscreen technology responds to a light tap of your finger
- Uses standard glass or AccuFet[®] electrodes with no adapters required

- Automatically corrects for temperature fluctuations with separate ATC probe in each channel
- Data acquisition via bidirectional communication with computer or printer
- Multiple language option: English, German, French, or Spanish built in
- Select from two procedural levels to match your requirements

An advanced design and incredible ease of use make the new Accumet AR50 the most versatile, reliable meter of its type you can buy today.

The AR50 offers you the advantages and versatility of having a research-grade pH, mV, ion, and conductivity meter, all in a single instrument, providing you with a dedicated system for a wide range of electrochemical analyses.



Features

- The largest display available; yet compact design saves valuable benchtop space
- User prompts and context-specific "HELP" screens easily guide user through operation
- Extensive setup screens allow you to customize the meter to your needs
- Autocalibrates with up to 5 pH buffers from any of 3 standard sets and 15 different buffers; also accepts custom buffers and manual calibration
- Innovative touchscreen makes it versatile and easy to use
- Dual channel meter allows you to measure two parameters at once and view results from both channels on screen simultaneously
- Reads pH to 0.001, mV to 0.1, plus conductivity, resistivity, salinity, and TDS
- Reads ISE in ppm, %, mg/mL, and mole/L
- · Direct and indirect ISE measurements, including known addition and subtraction, and analate addition and subtraction methods
- Extensive GLP capabilities with customizable printer or computer output and 250 reading datalog capability
- Print setup allows you to print exactly what you want, exactly when you want, not just a screen dump
- Time can be displayed in 12- or 24-hour formats and date in U.S. or international format
- Accepts standard BNC glass electrodes as well as an AccuFet electrode directly—no need for an adapter
- Accepts two-cell and four-cell conductivity probes
- Built-in replatinizing circuit
- Automatically corrects for temperature fluctuations with separate ATC (Automatic Temperature Compensation) probe in each channel

Simple customization lets you fit the system to your needs

You can easily customize the AR50 Meter screen to set measurement mode operating parameters, display only the information you need, or match your operating needs. There are two ways to do this:

- Choosing the "SYSTEM SETUP OPTIONS" menu lets you customize and simplify the screen displays to show exactly the parameters that suit your needs. You can edit the date, screen configuration, time, beeper status, and print configuration.
- · Customizing within each of the AR50's operating modes (pH/mV/ion/conductivity) lets you select

and set the operating parameters you wish to display, including resolution, buffer sets, and similar settings (different for each operating mode) to match your requirements.

You can also choose either of two procedural levels: Basic Level displays a reduced set of information and limits user access to the full list of setup parameters; Advanced Level allows user-configurable output and access to all setup parameters. Using this feature, scientists can configure the meter in Advanced Level and have a technician perform measurements in Basic Level, limiting the technician's ability to change setup parameters.

Specifications and Ordering Information	
Displays Screen size Measurement display height Temp./etc. display height	640 x 480 LCD 4¹/₂W x 6″H ³/₄″H ¹/₄″H
Memory Internal diagnostics Programmable data output Print interval Programmable alarm	250 data points Yes Output on stable, time, manual 1 to 9,999 sec. Yes
pH mode Range Resolution Relative accuracy Auto/manual buffer recognition Calibration points	-2.000 to +20.000 0.1/0.01/0.001 ±0.002 Yes 5
mV mode Range Resolution Accuracy	±1800.0 0.1 ±0.1
Ion mode Range Resolution Relative accuracy Calibration points Incremental methods	1 x 10 ⁻⁶ to 9.99 x 10 ¹⁰ 4 significant figures ±0.17n% 5 KA, KS, AA, AS
Conductivity mode Cell constants Ranges: Conductivity Resistivity Salinity Accuracy	0.1/1.0/10 0 to 3 x 10 ⁵ µS/cm 30megohm-cm to 100megohm-cm 2 to 42ppt ±0.5%
Temperature mode Range Resolution Accuracy Inputs/outputs	-5.0° to +105°C 0.1°C ±0.2°C
Meter size	2 BNC, 2 Pin, 2 ATC, 2-pin conductivity, bidirectional RS-232, 2 DIN (for FET and 4-cell conductivity)
Meter weight	6½W x 9¼L x 3½"H; 165 x 235 x 89mm 2.34 lb. (1.1kg)

Catalog No. **Description** 13-636-AR50 AR50 Meter Kit:

includes meter, Ag/AgCl combination pH electrode, ATC probe, electrode support arm and bracket, user manual, and power supply.

AR50B Meter Kit: 13-636-AR50B

includes meter, AccuTupH+® rugged bulb calomel combination pH electrode for Tris-buffer applications, ATC probe, electrode support arm and bracket, user manual, and power supply.

AR50A Meter only:

The Accumet® Research Model AR25 pH/mV/Ion Meter



Dual channels let you measure and view two parameters at once

- Powerful—measure pH and ion simultaneously
- Intuitive user interface makes it exceptionally easy to use
- Extra-large, scratch- and chemicalresistant LCD readout with dual display
- Reliable touchscreen technology responds to a light tap of your finger
- Automatically corrects for temperature fluctuations with separate ATC probe in each channel
- Data acquisition via bidirectional communication with computer or printer
- Multiple language option: English, German, French, and Spanish built in
- Uses standard glass or an AccuFet electrode with no adapter required

The Accumet AR25 Meter is fully equipped for the most demanding pH and ion measurements, yet is so simple to use almost anyone can do it. It offers Accumet accuracy and reliability plus innovative, dependable touchscreen technology, proven in thousands of applications worldwide. Its user-friendly design gives you an unparalleled range of user-determined setup, screen display, data storage, and printing options. Its dual-channel design is like having two meters in one. You can display one channel at a time, or toggle to split screen for simultaneous viewing. For the lab that needs the versatility of a research grade pH meter and an ion meter, it's an easy choice.

Features

- The largest display available; yet compact design saves valuable bench space
- User prompts and context-specific "HELP" screens easily guide user through operation
- Extensive setup screens allow you to customize the meter to your needs
- Autocalibrates with up to 5 pH buffers from any of 3 standard sets and 15 different buffers; also accepts custom buffers and manual calibration
- Innovative touchscreen makes it versatile and easy to use
- Dual channel meter allows you to measure two parameters at once, and view results from both channels on screen simultaneously
- Reads pH to 0.001, mV to 0.1
- Reads ISE in ppm, %, mg/mL, and mole/L
- Direct and indirect ISE measurements, including known addition and subtraction, and analate addition and subtraction methods
- Extensive GLP capabilities with customizable printer or computer output and 250 reading datalog capability
- Print setup allows you to print exactly what you want, when you want it, not just a screen dump
- Time can be displayed in 12- or 24-hour format, and date can be displayed in U.S. or international format
- Accepts standard glass electrodes or an AccuFet® electrode directly—no need for an adapter
- Automatically corrects for temperature fluctuations with ATC probe

Specifications and Ordering	Information
Displays	640 x 480 LCD
Screen size	41/2W x 6"H
Measurement display height	3/ ₄ "H
Temp./etc. display height	¹/₄″H
Menu options	Extensive
Memory	250 data points
Internal diagnostics	Yes
Programmable data output	Output on stable, time, manual
Print interval	1 to 9,999 sec.
Programmable alarm	Yes
pH mode	
Range	-2.000 to +20.000
Resolution	0.1/0.01/0.001
Relative accuracy	±0.002
Auto/manual buffer recognition	Yes
Calibration points	5
mV mode	4000
Range	±1800.0
Resolution	0.1 +0.1
Accuracy	±0.1
Ion mode Range	1 x 10 ⁻⁶ to 9.99 x 10 ¹⁰
Resolution	4 significant figures
Relative accuracy	+0.17n%
Calibration points	5
Incremental methods	KA, KS, AA, AS
Temperature mode	
Range	-5.0° to +105.0°C
Resolution	0.1°C
Accuracy	±0.2°C
Inputs/outputs	
	2 BNC, 2 Pin, 2 ATC,
	bidirectional RS-232, DIN (for FET)
Meter size	Dir (IOI I LI)
THE ICI SIZE	6 ¹ / ₂ W x 9 ¹ / ₄ L x 3 ¹ / ₂ "H
	165 x 235 x 89mm
Meter weight	
	2.34 lb. (1.1kg)
	(3)

Description Catalog No. AR25 Meter Kit: 13-636-AR25

includes meter, Ag/AgCl combination pH electrode, ATC probe, electrode support arm and bracket, user manual, and power supply.

AR25B Meter Kit: 13-636-AR25B

includes meter, AccuTupH+® rugged bulb Accu • pHast® combination pH electrode for Tris-buffer applications, ATC probe, electrode support arm and bracket, user manual and and power supply.

AR25A Meter only: 13-636-AR25A

The Accumet® Research Model AR20 pH/mV/Conductivity Meter



Same accuracy as our top model—pH, mV, and conductivity readout

- Large, easy-to-use LCD touchscreen displays information clearly
- Exceptional user interface with plain language prompts make it a breeze to use
- Select from two procedural levels to suit your needs
- Accepts 2-cell and 4-cell conductivity probes
- Built-in replatinizing circuit
- Automatically corrects for temperature fluctuations with ATC probe
- Data acquisition via bidirectional communication with computer or printer
- Multiple language option: English, German, French, or Spanish built in

The newly-designed Accumet AR20's powerful conductivity and pH modes provide the same performance and ease of use as our top-of-the-line AR50. You also get the convenience and benefits of our extra-large touchscreen technology, with an unequaled range of user-determined setup, screen display, data storage, and printing options.

Features

- The largest display available; yet compact design saves valuable bench space
- On-screen prompts in plain language and context-specific "HELP" screens easily guide user through operation
- Autocalibrates with up to 5 pH buffers from any of 3 standard sets and 15 different buffers; also accepts custom buffers, and manual calibration
- Innovative touchscreen makes it versatile and easy to use
- Reads pH to 0.001, mV to 0.1, plus conductivity, resistivity, salinity, and TDS
- Extensive GLP capabilities with customizable printer or computer output and 250 reading datalog capability
- Autoranging capability
- Print setup allows you to print exactly what you want, when you want, not just a screen dump
- Time can be displayed in 12- or 24-hour format; date can be displayed in U.S. or international formats
- Accepts standard glass electrode or an AccuFet® electrode directly—no need for an adapter

On a disasting and Outpuin	Information
Specifications and Ordering	
Displays Carean size	640 x 480 LCD
Screen size Measurement display height	4¹/₂W x 6″H ³/₄″H
Temp./etc. display height	1/4"H
1. 1.	* *
Memory Internal diagnostics	250 data points Yes
Programmable data output	Output on stable,
r rogrammable data output	time, manual
Print interval	1 to 9,999 sec.
Programmable alarm	Yes
pH mode	
Range	-2.000 to +20.000
Resolution	0.1/0.01/0.001
Relative accuracy	±0.002
Auto/manual buffer recognition	Yes
Calibration points	5
mV mode	
Range	±1800.0
Resolution	0.1
Accuracy	±0.1
Conductivity mode	
Cell constants	0.1/1.0/10
Ranges:	0 to 0 v 105vC/om
Conductivity Resistivity	0 to 3 x 10⁵µS/cm 30megohm-cm to
nesistivity	100megohm-cm
Salinity	2 to 42ppt
Accuracy	0.5%
Temperature mode	
Range	-5.0° to +105.0°C
Resolution	0.1°C
Accuracy	±0.2°C
Inputs/outputs	
	BNC, Pin, ATC, 2-pin
	conductivity, bidirectional RS-232, 2 DIN (for FET
	and 4-cell conductivity)
Meter size	
	6 ¹ / ₂ W x 9 ¹ / ₄ L x 3 ¹ / ₂ "H
	165 x 235 x 89mm
Meter weight	
	2.34 lb. (1.1kg)
Description	Catalog No.

Description Catalog No. AR20 Meter Kit: 13-636-AR20

includes meter, Ag/AgCl combination pH electrode, ATC probe, electrode support arm and bracket, user manual, and power supply.

AR20B Meter Kit: 13-636-AR20B

includes meter, AccuTupH+® rugged bulb Accu*pHast® combination pH electrode for Tris-buffer applications, ATC probe, electrode support arm and bracket, user manual and power supply.

AR20A Meter only: 13-636-AR20A

The Accumet® Research Model AR15 pH/mV/°C Meter



An easy-to-use, dedicated pH/mV meter for research-grade measurements

- Large, scratch- and chemicalresistant LCD readout with reliable touchscreen technology for ease of operation
- Intuitive operation with on-screen, plain language prompts and "HELP" screens that guide you through the measurement process
- Data acquisition via bidirectional communication with computer or printer
- Multiple language option: English, German, French, or Spanish built in

The Accumet AR15 Meter measures pH and mV with all the power and ease of use of the top-of-the-line AR50. It automatically recognizes USA, Euro, NIST, or custom buffers. Its remarkable user interface makes it exceptionally easy to use, and a full complement of plain-language prompts and "HELP" screens quide you through the measurement process.

Features

- Innovative, reliable touchscreen technology makes it versatile and easy to use
- User prompts and context-specific "HELP" screens easily guide user through operation
- Extensive setup screen allows you to customize the meter to your needs
- Autocalibrates up to 5 pH buffers from any of 3 standard sets and 15 different buffers; also accepts custom buffers and manual calibration
- Reads pH to 0.001 and mV to 0.1
- Extensive GLP capabilities with customizable printer or computer output and 250 reading datalog capability
- Print setup allows you to print exactly what you want, when you want it, not just a screen dump
- Time can be displayed in 12- or 24-hour format; date can be displayed in U.S. or international format
- Accepts standard glass electrode or AccuFet® electrode directly—no need for an adapter
- Automatically corrects for temperature fluctuations with ATC probe

Displays	640 x 480 LCD
Screen size	41/2W x 6"H
Measurement display height	3/4"H
Temp./etc. display height	¹/ ₄ ″H
Memory	250 data points
Internal diagnostics	Yes
Programmable data output	Output on stable, time, manual
Print interval	1 to 9,999 sec.
Programmable alarm	Yes
pH mode	
Range	-2.000 to +20.000
Resolution	0.1/0.01/0.001
Relative accuracy	±0.002
Auto/manual buffer recognition	
Calibration points	5
mV mode	
Range	±1800.0
Resolution	0.1
Accuracy	±0.1
Temperature mode	
Range	-5.0° to +105.0°C
Resolution	0.1°C
Accuracy	±0.2°C
Inputs/outputs	
	BNC, Pin, ATC, bidire tional RS-232, DIN (for FET)
Meter size	
	6 ¹ / ₂ W x 9 ¹ / ₄ L x 3 ¹ / ₂ "H 165 x 235 x 89mm
Meter weight	
	2.34 lb. (1.1kg)
Description	Cotolog No.

Description Catalog No. AR15 Meter Kit: 13-636-AR15

includes meter, Ag/AgCl combination pH electrode, ATC probe, electrode support arm and bracket, user manual, and power supply.

AR15B Meter Kit: 13-636-AR15B

includes meter, AccuTupH® rugged bulb double-junction combination pH electrode for Tris-buffer applications, ATC probe, electrode support arm and bracket, user manual, and power supply.

AR15A Meter only: 13-636-AR15A

The Accumet® Research Model AR 10 pH/mV/°C Meter



Performance, reliability, and economy in an easy-to-use digital pH meter

- Simple design and operation plus research-grade performance
- Sleek, modern design
- Compact—small footprint saves valuable bench space

The Accumet AR10 Meter gives you reliable, research-grade pH and mV measuring capabilities in a package that is ideal for classrooms and industrial labs. A big, four-digit LCD makes reading measurements easy, and four-knob rotary control makes it simple to use.

Features

- Reads pH to 0.01 and mV to 1
- Solid-state circuitry with digital display and rotary knobs
- Ideal for educational use
- Automatically corrects for temperature fluctuations with ATC probe

Replacement Parts and Accessories for Accumet AR Series Benchtop Meters

Description	Catalog No.
For Accumet AR Series Benchtop Meters	
pH combination electrode, Ag/AgCI, single junction, glass body, BNC connector; replacement for AR10, AR15, AR20, AR25, AR50	13-620-285
AccuFet® solid-state pH/ATC combination electrode, Ag/AgCl	13-620-755
ATC probe, stainless steel; for all AR Series meters	13-620-19
Electrode support arm for all AR Series meters	13-637-671
Electrode arm bracket for all AR Series meters	13-637-671A
Power Supply—115V, 60Hz, US plug	13-636-100
For Accumet AR50, AR25, AR20, AR15 Benchtop Meters	
Printer, including printer, cable, ribbon, paper. For 115V	13-637-670
Printer Paper	13-637-669
Printer Replacement Ribbon	13-637-668
Printer Cable	13-637-667
Computer Cable for AR Series (9-pin to 9-pin)	13-637-680

Conductivity Cells for AB30, AR20, AR50				
	2-Cell Cond	luctivity Cell	4-Cell Cond	uctivity Cell
	Glass Body	Epoxy Body	Glass Body	Epoxy Body
Accumet Immersion Type Conductivity Electrodes				
Cell constant 0.1cm ⁻¹	13-620-156	13-620-161		
Cell constant 1.0cm ⁻¹	13-620-155	13-620-160	13-620-163	13-620-165
Cell constant 10.0cm ⁻¹	13-620-157	13-620-162	13-620-164	13-620-166

Specifications and Orderin	g Information
Displays	4-digit LCD
Screen size	21/ ₄ W x 1"H
Measurement display height	3/ ₄ "H
Keypad controls Programmable data output	4-knob rotary Recorder (±1800mV)
pH mode	Tiodordor (±1000mv)
Range	-1.99 to +19.99
Resolution	0.01
Relative accuracy	±0.02
Manual buffer recognition	Yes
Calibration points	2
mV mode	
Range	±1800.0
Resolution	1
Accuracy	±1
Temperature mode	0001 100 000
Range Resolution	0°C to 100.0°C 0.1°C
Accuracy	+0.5°C
Inputs/outputs	±0.5 €
ilipuis/oulpuis	BNC, Pin, ATC,
	Recorder, DIN
	(for FET)
Meter size	
	6 ¹ / ₂ W x 9 ¹ / ₄ L x 3 ¹ / ₂ "H 165 x 235 x 89mm
Meter weight	
	2.34 lb. (1kg)
Description	Catalog No.
AR10 Meter Kit:	13-636-AR10

probe, electrode support arm and bracket, user manual, and power supply.

AR10A Meter only:

13-636-AR10A

The Accumet® Basic Model **AB30** Conductivity/°C Meter



An accurate, affordable, easy-to-operate dedicated conductivity meter

- Easy-to-read custom LCD—displays information clearly
- Shows measurements in conductivity, resistivity, or total dissolved solids units
- Clear user interface with plain language prompts make it a breeze

The Accumet Basic AB30 Conductivity Meter features a large digital display that shows stability indicators as well as user prompts and error messages in plain language, making it exceptionally easy to use. The AB30 Meter measures conductivity, resistivity, and total dissolved solids, and all measurements are corrected for temperature fluctuations with the ATC probe (supplied with kit).

Features

- Reads conductivity, resistivity, and total dissolved solids to 4 significant figures
- Large display shows measurement and temperature at all times
- Easy-to-use operation with user prompts on screen, stability indicators, and error messages
- Automatically corrects for temperature fluctuations with ATC probe
- Accepts two-cell and four-cell conductivity probes
- Small, compact size conserves valuable bench space

Replacement Parts and Accessories for Accumet AB Series Benchtop Meters

Description	Catalog No.
pH/ATC electrode, Ag/AgCl, single junction, polypropylene, BNC connector; replacement for AB15	13-620-530
pH/ATC electrode, calomel, single junction, polypropylene, BNC connector; replacement for BioBasic AB15B	13-620-531
AccuFet® solid-state pH/ATC combination electrode, Ag/AgCl	13-620-755
ATC probe, stainless steel; for all AB Series meters	13-620-19
Electrode support arm for all AB Series meters	13-637-671
Electrode arm bracket for all AB Series meters	13-637-671A
Power Supply—115V, 60Hz, US plug	13-636-100

Conductivity Cells for	AB30			
	2-Cell Cond	luctivity Cell	4-Cell Cond	uctivity Cell
	Glass Body	Epoxy Body	Glass Body	Epoxy Body
Accumet Immersion	Type Conductivity	Electrodes		
Cell constant 0.1cm ⁻¹	13-620-156	13-620-161		
Cell constant 1.0cm ⁻¹	13-620-155	13-620-160	13-620-163	13-620-165
Cell constant 10.0cm ⁻¹	13-620-157	13-620-162	13-620-164	13-620-166

Specifications and Orderin	g Information
Display	Custom LCD
Screen size	3W x 4 ¹ / ₄ "H
Measurement display height	3/4"H
Temp./etc. display height	5/ ₁₆ "H
Keypad controls	5-key soft touch
	membrane
Conductivity mode	
Cell constants	0.1, 1.0, 10
Ranges:	
Conductivity	0 to 3 x 10⁵µS/cm
Resistivity	30megohm-cm to
TDC	100megohm-cm
TDS	0 to 1.8 x 10⁵ppm +0.5%
Accuracy	±0.5%
Temperature mode	5.001 405.000
Range	-5.0° to +105.0°C
Resolution	0.1°C +0.5°C
Accuracy	±0.5°C
Inputs	0
	2-cell (2 pins) con- ductivity, 4-cell (DIN)
	conductivity, ATC probe
Meter size	
	5 ¹ / ₂ W x 7 ¹ / ₂ L x 3 ¹ / ₄ "H
	140 x 191 x 83mm
Meter weight	
	1.86 lb. (0.8kg)
Description	Catalog No.
AB30 Meter Kit:	13-636-AB30
includes meter, ATC probe, electoracket, user manual, and power	ctrode support arm and er supply.
AB30A Meter only:	13-636-AB30A
includes meter, electrode suppo	ort arm and bracket, user

The Accumet® Basic Model AB15 pH/mV/°C Meter



Intuitive, simple operation, and high accuracy in a compact, affordable meter

- Same quality as our top-of-the-line AR50 Research Meter
- Largest LCD readout in its class
- Accepts standard glass or AccuFet® electrode directly with no adapter needed

Easy operation, accuracy, and affordability set the Accumet Basic AB15 Meter apart from the competition.

A large custom LCD makes the AB15 Meter easy to read, while plain language prompts and error messages guide the user through measurements. A rugged, chemical-resistant fivebutton soft touch membrane keypad controls all operations. The AB15 Meter automatically corrects for temperature fluctuations using an ATC probe.

The AB15 Meter Kit includes Ag/AgCl combination pH/ATC electrode. Life science and other researchers using Tris buffers can choose the AB15B BioBasic Meter Kit, complete with calomel electrode with built-in ATC probe.

For laboratories needing a basic, accurate, easy-to-use pH/mV meter, the AB15 Meter is the perfect choice.

Features

- Measures pH, mV, and relative mV to 0.01 pH and 1mV
- Large display shows pH or mV and temperature at all times
- Autocalibrates with up to 5 pH buffers from any 3 standard sets and 15 different buffers
- Easy-to-use operation with user prompts, stability indicators, and error messages on screen
- Accepts a standard glass electrode or AccuFet® electrode directly—no adapter needed
- Automatically corrects for temperature fluctuations with ATC probe
- Standardization curves for standard glass and the AccuFet electrode stored in memory—switch electrodes quickly with no need to restandardize
- Small, compact size conserves valuable bench space

Specifications and Orderin	g Information
Display	Custom LCD
Screen size	3W x 4 ¹ / ₄ "H
Measurement display height	³ / ₄ "H
Temp./etc. display height Keypad controls	⁵ / ₁₆ "H 5-key soft touch
Neypau controls	membrane
Internal diagnostics	
	Yes
pH mode	
Range	-1.99 to +19.99
Resolution	0.1/0.01 +0.01
Relative accuracy Automatic buffer recognition	±u.u i Yes
Calibration points	5
mV mode	
Range	±1800.0
Resolution	1
Accuracy	±0.2
Temperature mode	
Range	-5.0°C to +105.0°C
Resolution	0.1°C
Accuracy	±0.2°C
Inputs	
	BNC, Pin, ATC, DIN (for FET)
Meter size	
	5 ¹ / ₂ W x 7 ¹ / ₂ L x 3 ¹ / ₄ "H 140 x 191 x 83mm
Meter weight	
	1.86 lb. (0.8kg)

Description Catalog No. AB15 Meter Kit: 13-636-AB15

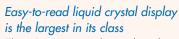
includes meter, Ag/AgCl combination pH/ATC electrode, electrode support arm and bracket, user manual, and power supply.

AB15B BioBasic Meter Kit: 13-636-AB15B

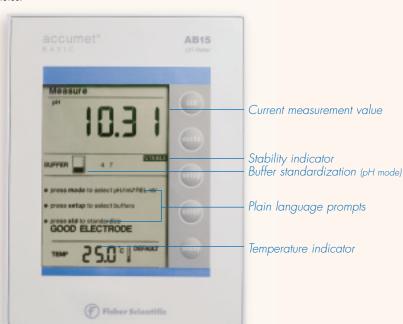
includes meter, calomel combination pH/ATC electrode for Tris-buffer applications, electrode support arm and bracket, user manual, and power supply.

AB15A Meter only: 13-636-AB15A

includes meter, electrode support arm and bracket, user manual, and power supply.



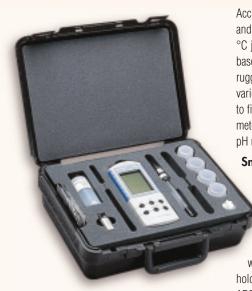
The AB15 Meter's large, liquid crystal display (LCD) makes it exceptionally easy to use and read. Operation is controlled by only five soft touch membrane keys.



Accumet® Portable Meters pH/mV/lon/°C



Accurate, portable, easy to use. Precise measurements any time, anywhere



Accumet Portable (AP) pH meters are so compact and easy to use, you can test for pH, mV, lon, and °C just about anywhere. Their microprocessorbased accuracy and precision, coupled with their rugged, waterproof design make them ideal for a variety of remote testing applications. Offering up to five calibration points, Accumet Portable pH meters provide the ultimate in portable pH measurement.

Small enough to fit in your shirt pocket

Compact, rugged Accumet Portable pH meters can be used almost anywhere, from the R&D lab to the factory floor to the job site. Their trim shape and rubberized, wraparound grips make them comfortable to hold and operate with just one hand. Choose the AP61 for pH-only measurements, the AP62 for pH/mV, or the AP63 for pH, mV, or ion applications. All models also measure temperature in °C.

Easy to use, comfortable to hold

A user interface similar to that used in Accumet Basic Benchtop meters makes Accumet Portable pH meters exceptionally easy to operate. Six sealed, soft-touch entry keys let you scroll easily through menu options. pH mode features selectable resolution to 0.01, automatic temperature compensation, automatic recognition of 5 buffers, and five-point calibration. (The AP63 also features automatic recognition of 5 standards and five-point calibration in ion mode.) Plain language onscreen prompts and error messages make getting good results easy, even for inexperienced operators.

Rugged and waterproof

A tough, ABS plastic housing lets Accumet Portable pH meters take hard use in stride. In fact, to help you handle even the most difficult remote measurement applications, they're waterproof to IP67 standards and will even float if dropped in water. Accumet Portable pH meters operate from a 9V battery or with an optional AC adapter.

Backed by a full two-year warranty

Not only are Accumet Portable meters accurate, versatile, and easy to use, they're backed by a full two-year warranty. Should any Accumet meter fail to perform to specification under normal use during the first two years, it will be repaired or replaced at no charge.

Three models to match your remote testing and sampling requirements

- Large, easy-to-read LCD readouts
- Same simple, intuitive operation as Accumet Basic Series Benchtop Meters
- Six-button, soft touch keypad
- Ergonomic design makes them comfortable to hold
- Waterproof to IP67 standards in fact, they float!

Replacement Parts and Accessories for Accumet AP60 Series Portable Meters

Description	Catalog No.
Power Supply, 115V, 60Hz, US plug	13-636-100
pH/ATC electrode, Ag/AgCl, single junction, polypropylene, refillable	13-620-AP50
pH/ATC electrode, calomel, single junction, polypropylene, refillable	13-620-AP51
pH/ATC electrode, double junction, epoxy body, gelled Ag/AgCl electrolyte	13-620-AP52
ATC probe, stainless steel for all AP60 Series meters	13-620-AP53
Accumet Portable Lab (Carrying Case)	13-636-AP69

Note: Accumet AP Series Portable Meters are only waterproof to IP67 standards when used with AP Series electrodes above, and with AC adapter cap in place. Meters are not waterproof when used with AC adapter.

pHree Trial Offer makes it so easy to try one

We're so sure you'll agree that Accumet®
Portable meters offer the best performance,
versatility, and value available that we invite
you to try one FREE for 30 days with no risk
and no obligation. To schedule your Accumet
pHree Trial, contact your Fisher
Sales Representative.



Specifications and Orderi	ng Information
Accumet Portable AP61 p	
Displays	Custom LCD
Screen size	115/16W x 21/2"H
Measurement characters	⁷ / ₁₆ "H
Other data characters	1/4"H
Keypad controls	6-key membrane
Internal diagnostics	Self-test, error message
Waterproofing	IP67 [†]
pH mode	
Range	-1.99 to +19.99
Resolution	0.1/0.01
Relative accuracy	±0.01
Auto buffer recognition Calibration points	2, 4, 7, 10, 12 1 to 5
	1 10 0
Temperature mode Range	-5.0° to +100.0°C
Resolution	0.1°C
Relative accuracy	+0.3°C
Compensation	Automatic
Inputs	Automatio
проіз	BNC. ATC
Electrical requirements	DNO, ATO
Liechical requirements	9V battery or optional
	AC adapter
Battery life	
	>200 hr.
Description	Catalog No.

, i	AP61 Meter Kit‡ AP61A Meter only‡	13-636-AP61 13-636-AP61A
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Specifications and Ordering Information

Accumet Portable AP62 pH/mV/°C Meter

Same specifications as Model AP61 plus:

mV mode Bange

 $\begin{array}{ll} \text{Range} & 0 \text{ to ± 999} \\ \text{Resolution} & 1.0 \\ \text{Relative accuracy} & \pm 0.1 \end{array}$

Description	Catalog No.
AP62 Meter Kit‡	13-636-AP62
AP62A Meter only‡	13-636-AP62A

Specifications and Ordering Information

Accumet Portable AP63 pH/mV/lon/°C Meter

Same specifications as Model AP62 plus:

Ion mode

 $\begin{array}{lll} \mbox{Range} & 1 \times 10^3 \mbox{ to } 9.99 \times 10^4 \\ \mbox{Relative accuracy} & \pm 0.17 \mbox{n}\% \\ \mbox{Calibration points} & 5 \\ \mbox{Resolution} & 2 \mbox{ or } 3 \mbox{ significant digits} \\ \end{array}$

 Description
 Catalog No.

 AP63 Meter Kit‡
 13-636-AP63

 AP63A Meter only‡
 13-636-AP63A

†**IP67 Standard:** Meter can be submerged for 30 minutes to a depth of 1m and still remain operational.

‡Meter only versions include meter, 9V battery, and user manual.

Meter Kits include meter, 9V battery, Accumet liquid-filled polymer body pH/ATC Ag/AgCl electrode, user manual and hard carrying case with 4 empty buffer bottles and 4 single-use pH buffer pouches.

Electrochemistry Theory and Measurements

pH Theory and Measurement pH Working Theory

Since its 1909 introduction, pH measurement has become increasingly important in laboratories and industries. Close pH control is of primary importance in life processes, in sanitary engineering and electroplating, and in textile, pharmaceutical, and food industries. Modern meters and new electrodes have made pH measurement as simple and convenient as temperature measurement.

In general, pH is a measure of the degree of acidity or alkalinity of a substance. It's related to the effective or active acid concentration of a solution by this equation:

$$pH = -log\alpha H^+$$

with aH* representing the activity of the hydrogen ions in the solution. Neglecting activity effects, the equation above reduces to:

$$pH = -log[H^+]$$

with [H*] representing the concentration of hydrogen ions in solution. pH is sometimes referred to as the power of the hydrogen ion in solution. The pH of the strong acid 0.01 molar HCl is equal to 2, since its hydrogen ion concentration is 10⁻² molar:

$$pH = -log[10^{-2}]$$

 $pH = 2$

The pH scale in water is usually considered to range from 0 to 14, or from an active acid concentration of $[1 \times 10^{\circ}]$ to $[1 \times 10^{-14}]$ (1.0M to 0.0000000000000001M). Some current pH meters can measure to -2pH (10° M) and to +20pH (10° 0M).

The pH scale is based on the dissociation constant of water. In pure water, a very small number of molecules react with one another to form hydronium ions (H_3O^*)—which account for acidic properties—and hydroxide ions (OH *)—which account for the basic properties of an aqueous solution.

$$2H_2O \Leftrightarrow H_3O^+ + OH^-$$

or, for simplicity:

$$H_2O \Leftrightarrow H^+ + OH^-$$

At 25°C, pure water dissociates until the acid [H $^{+}$] and base [OH $^{-}$] concentrations are equal, at 1 x 10 $^{-7}$ molar. The product of both concentrations is the dissociation constant K_{w} :

$$\begin{split} K_w &= [H^+] [0H^-] \\ K_w &= [1 \times 10^{-7}] [1 \times 10^{-7}] \\ K_w &= [1 \times 10^{-14}] \end{split}$$

Because its hydrogen ion concentration $[H^*]$ equals 1 x 10^{-7} molar, the pH of pure

water at 25°C is 7. This is referred to as the neutrality point.

In aqueous solutions, at 25° C, the product of [H+] and [OH-] must remain constant at 1 x 10^{-14} ; an increase in either acid or base concentration always results in a decrease in the other. Hence, a solution of the strong base 0.01M NaOH will have a hydrogen ion concentration of:

$$H^+ = \frac{K_W}{[0H^-]} = \frac{10^{-14}}{10^{-2}} = 10^{-12}$$

Thus:

$$pH = -log [10^{-12}]$$

 $pH = 12$

pH Measurement Systems

Although pH can be determined using colorimetric indicators or indicator papers, the preferred method is potentiometric measurement, using a pH meter and electrode system. This method is based on the fact that certain electrodes, immersed in a solution, produce a voltage that is related to the pH value of the solution in a very precise way. This voltage can be predicted by the Nernst equation. Simplified, the equation is:

$$E_{meas} = E^0 - \frac{2.3}{nF} RT (pH)$$

where E_{meas} is the measured voltage; E^0 is the total of all constant voltages in the measurement system; R is the Gas Law constant; T is the temperature in ${}^{\circ}K$; n is the charge of the ion; and F is Faraday's constant.

The pH Meter

The pH Meter is a specialized voltmeter. It can accurately measure small voltage changes at extremely high resistances exhibited by electrodes. And it can adjust to the pH and voltage characteristics of the electrode system. Most modern pH meters incorporate manual or automatic temperature compensation to correct for variations in pH value of a given solution with sample temperature, as expressed in the Nernst equation. To compensate for electrode output variations, most meters are equipped with a slope or efficiency control to adjust the meter to match electrode voltage exactly. Microprocessor-based meters are programmed to solve the Nernst equation, taking into account electrode voltage, efficiency and temperature. Most provide automatic buffer recognition and standardization, as well as user prompting, error messages and diagnostic circuitry to simplify operation and reduce error.

Standardization Buffers

Buffers—solutions of known pH value—adjust the system to display accurate measurements. Buffers are available as ready-to-use or concentrated solutions, in capsules and as prepackaged salts.

All have the special characteristic of **resisting pH change** upon dilution or acid/base contamination. For best accuracy, a two-point standardization is per-

formed: first with a buffer value close to the electrode system's zero potential (typically pH 7); and next with an acid or base buffer whose value brackets the expected pH value of the sample. Microprocessor-based meters may permit additional calibrations—up to five points in some models—for increased definition of electrode slope, allowing measurement of samples with widely varying pH values. For best accuracy, all buffers should be at the same temperature as the sample.

The following table lists three buffers established by the National Institute of Standards and Technology, with their pH values at various temperatures.

Temp.	4.01 Buffer 0.05M Potassium Acid Phthalate	6.86 Buffer 1.025M KH ₂ PO ₄ 0.025M Na ₂ HPO ₄	9.18 Buffer 0.01M Borax
0°C	4.00pH	6.98pH	9.46pH
10°	4.00	6.92	9.33
20°	4.00	6.88	9.22
25°	4.01	6.86	9.18
30°	4.02	6.85	9.14
40°	4.04	6.84	9.07
50°	4.06	6.83	9.01
60°	4.09	6.84	8.96
70°	4.13	6.85	8.92
80°	4.16	6.86	8.88
90°	4.21	6.88	8.85

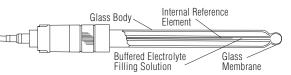
The Electrode System

The electrode system consists of two half cells: a pH indicating electrode, which develops a potential dependent on the pH of a solution; and a reference electrode, which provides a constant potential and completes the electrical circuit.

Using separate pH indicating and reference half cells lets you select each cell without compromise, tailor the system precisely to the sample's nature and achieve the best accuracy. It can mean lower replacement costs, too, since usually only one of a pair is broken.

Nevertheless, the combination electrode—an indicating half cell and a reference half cell joined coaxially—is being used more frequently, for the convenience and compactness it offers. Those which incorporate a temperature probe can provide temperature readout and compensation with ATC-capable meters. Some types of pH, reference and combination electrodes are discussed below and on the following page.

The Electrode Pair



pH Indicating Electrodes

The traditional pH electrode consists of a glass sensing membrane of special electrically-conductive glass, bonded to a nonconductive tube of glass or plastic, called the body; and an internal reference element—usually Ag/AgCl—immersed in a buffered electrolyte of fixed pH value and ionic concentration.

This design assures that constant potentials are developed on the inner surface of the glass membrane, and on the internal reference element. When the electrode is immersed in a solution of pH 7, the sum of these fixed voltages approximately balances the voltage developed on the outer surface of the glass membrane and the separate reference electrode. Thus, in a pH 7 solution, the total potential output of the system is near OmV. In solutions of more or less than pH 7, the potential on the outer membrane surface changes in proportion to the sample pH. The voltage change is sensed by the meter and displayed as a pH value.



Recent designs have replaced the glass pH membrane with a sensor comprising the gate connection of an ion-sensitive FET (Field Effect Transistor). A pH-related potential is developed across the gate of this semiconductor, which, in turn, controls the current flowing through the transistor. The current output is in essence dependent on the activity of the hydronium ion in the solution being measured. Most such electrodes are designed as combination electrodes. (See pp. 19 and 22 for Accufete model.)

Selecting a pH Indicating Electrode

Fisher offers glass pH indicating electrodes with membranes of two types of glass. Both types are usable over the 0 to 14pH range, and feature low error in samples with high sodium content. (See pp. 19 and 22 for specifications and ordering information.)

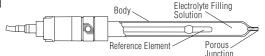
FS-5 Universal Glass Membranes have low electrical resistance—less than 100 megohms at 25°C. They're recommended for all routine pH and titration measurements, as well as measurements at low temperatures. Their sodium error is less than 0.01pH (in 0.1N NaOH at 25°C). New models with an extra-thick membrane are available; their rugged bulbs are 40mil in thickness, compared to the standard thickness of 8mil, making them up to five times more durable under rough handling.

Full Range High pH Glass Membranes have somewhat higher electrical resistance plus excellent chemical durability. They're ideal for frequent or long-term measurements in solutions of very high or very low pH value, and feature sodium ion error of less than 0.03pH (in 0.1N NaOH at 25°C).

As an alternative to the glass bulb, Fisher offers AccuFet® combination electrodes with built-in ATC probe and ion-selective field effect transistor sensor. The solid-state sensor precludes breakage, since there's no glass bulb, and the body is of durable epoxy.

(See p. 22 for ordering information.)

Reference Electrodes



The basic requirements of the reference half cell are to complete the electrical circuit and to provide a stable reference potential. The probe contains a reference element—usually calomel or silver/silver chloride—immersed in an electrolyte solution of fixed ionic concentration. This produces a constant voltage, despite sample composition. The circuit is completed by allowing a small flow of the electrolyte to pass through a junction in the probe tip; or by ionic diffusion in an electrode filled with gelled electrolyte.

Selecting a Reference Electrode

To maintain constant junction potential under specific measurement conditions requires an adequate flow of electrolyte. So the choice of reference electrode is made based on the electrode's flowrate—which depends on its junction type—and on the characteristics of its internal element and electrolyte relative to the sample.

Liquid Junctions cover the majority of applications, with flowrates from 0.5 to $100\mu L/hr$. These junctions perform best with the least junction potential. In most cases, this occurs with a fast-flowing junction; but rapid electrolyte flow can cause significant sample contamination. Hence the need for a variety of junctions to provide a selection of flowrates.

- Sleeve and annular junctions provide flowrates—
 to 100µL/hr.—for enhanced conductivity and
 response in solutions of low ionic strength, such as
 deionized water, and in difficult samples such as
 slurries, emulsions and suspensions. Sleeve junctions can be removed or slid out of the way to clean
 or refresh the junction. The reverse sleeve type is
 designed to prevent loss of the sleeve. Some annular junction models have spring-loaded caps for
 flow adjustment and easy cleaning.
- Porous ceramic junctions offer moderate flowrates of about 8µL/hr., providing excellent solution contact with minimal need for electrolyte replenishment. They're recommended for the majority of routine pH and titration measurements.
- Cracked bead junctions allow the least amount of electrolyte flow—only 0.5 to 5.0µL/hr.—yet exhibit minimal problems with clogging of the junction.

Nonflowing junctions are most often found in polymer body electrodes designed for applications where minimal electrode maintenance is desirable. The electrolyte is usually a gel; contact is by ionic diffusion through a porous junction (rather than by liquid flow). These types are useful for in-plant monitoring, field measurement and classroom use.

Double-junction electrodes permit isolation of the reference element and its electrolyte by enclosing them in an inner electrode body. The flowrate is determined by the second junction, contained in the outer body of the electrode. A choice of outer junctions and electrolytes permits matching of electrode to sample.

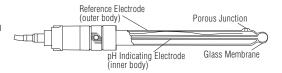
Internal elements provide a fixed potential of excellent stability. The choice in most applications is between calomel (Hg/Hg₂Cl₂) and silver/silver chloride (Ag/AgCl) elements. Selection is based primarily on sample temperature, but the effects of the electrolyte solution on the sample should also be considered.

 Calomel (mercury/mercurous chloride) internals are recommended for most routine applications. The electrolyte is a saturated solution of potassium chloride (KCI), so they're easy to use and maintain. Typical temperature range is -5° to +80°C.

• Silver/silver chloride internals

have a wider temperature range of -5° to +100° or 110°C. The electrolyte is 4M KCI, saturated with AgCI to prevent dissolution of the silver chloride portion of the internal element. Though Ag/AgCI internals are usable at higher temperatures, the presence of AgCI in the electrolyte may cause junction clogging with low-conductivity samples, and certain silver-sensitive samples—such as biomedical specimens—can react with the electrolyte, causing additional measuring difficulties. Double-junction Ag/AgCI electrodes are available to prevent sample contamination by heavy metals and reduce junction clogging as well. (See pp. 22 and 23 for Accumeter reference electrodes.)

The Combination Electrode



The combination electrode contains both pH-sensing and reference half cells, joined in a single body. In most combinations, the pH-sensing cell is enclosed in a tube; its membrane is exposed at the probe tip. The annular space around the inner tube contains the reference element, electrolyte and junction. Advantages of this design are handling convenience and compactness—allowing measurements of smaller samples and in narrow vessels.

They're practical for field measurements too, and come in a wide range of lengths and diameters for use in a variety of containers: test tubes, centrifuge tubes, fraction collectors, cuvets and flasks among others. Even greater compactness is achieved in models which incorporate an Automatic Temperature Compensation probe (ATC) within the combination electrode's body. (See pp. 21 and 22 for Accumet combination electrodes, including Accu • pHast® variable temperature and AccuFet solid-state electrodes.)

Electrochemistry Theory and Measurements (cont.)

Conductivity Theory and Measurement

Conductivity Theory

Conductance is a value associated with the ability of primarily aqueous solutions to carry an electrical current. Though water itself is a rather poor conductor of electricity, the presence of ions in an aqueous solution increases the solution's conductance considerably. The current is carried by migration of ions dissolved in the water. The solution's conductance is proportional to the concentration of ionic species present, as well as the ions' charge and mobility, and is defined as the reciprocal of the solution's electrical resistance:

Conductance = 1/Resistance

Conductivity measurements, however, generally involve determining the resistance of a small portion of solution between two parallel electrodes when an alternating potential is applied. Conductivity values are related to the conductance of a solution by the physical dimensions—area and length—of the measuring cell, called the cell constant.

Conductance = Conductivity x Cell Constant Cell Constant = Area/Length

Polarization effects—which would impair the accuracy of the measurements—are avoided by using AC potential, rather than DC; and by coating the electrode's surface area with platinum black. Conductivity measurements provide an easy way to assess the quality of water or of various aqueous solutions.

Conductivity measurements are reported as Siemens/cm, since the value is measured between opposite faces of a cell of known cubic configuration. With aqueous solutions, values are most frequently measured in μ Siemens per cm (μ S/cm) or milliSiemens per cm (μ S/cm).

Some users prefer to use resistivity units to describe their water, particularly where very pure water is involved. Units used to measure resistivity are megohms-cm; they're simply the reciprocals of the conductivity units. The chart below shows the relationship between these two units.

Resistivity megohms-cm
18.0
10.0
1.0
0.4
0.1

Conductivity Measurement

Conductivity Cells, such as the Accumet models on p. 24, consist of a glass or epoxy body in which platinum or platinized sensing elements are fixed. These sensors contact the solution whose conductivity value is sought. Cell constants—each cell's unique geometric configuration—are determined by measuring the conductivity of a standard of known conductive value, typically an aqueous solution of KCI.

To produce a resistance signal appropriate for a given conductivity meter, it's important to choose a cell with a compatible cell constant. The following table shows optimum conductivity ranges for cells of three different constants.

Cell Constant	Optimum Conductivity Range (µS/cm)
0.1	0.5 to 200
1.0	10 to 2000
10.0	1000 to 200,000

Before use, cells should be conditioned in distilled water for at least 10 minutes, then standardized to determine the cell constant, following instructions supplied with the meter. Conductivity changes considerably with changes in temperature, but the changes are fairly linear, and modern meters provide compensation for accurate standardization. (See p. 24 for Accumet® conductivity cells.)

Metallic Electrodes

ORP Theory

Oxidation-reduction measurements are used to determine the oxidizing or reducing properties of solutions. to monitor chemical reactions, in quantitative determinations of ions and to detect endpoints in titrations. In chemical oxidation-reduction reactions, electrons are transferred from one species—the reducing agent—to another—the oxidizing agent. Substances vary tremendously in their tendency to act as either oxidizing or reducing agents. Tables of standard oxidation-reduction potentials in the literature are determined relative to the Standard Hydrogen Electrode (SHE) which has a potential of 0.00V at all temperatures (with hydrogen ion activity at 1.00 and the partial pressure of hydrogen gas at 1.00 atmosphere). The standard potential (E⁰) of any oxidationreduction refers to the potential developed vs. the SHE, when all species are at an activity of 1.00 (unit activity). A general equation for a reduction reaction is:

$$0x + ne^{-} = Red$$

where n is the number of electrons, Ox is the activity of the oxidized form and Red is the activity of the reduced form.

The ORP can be expressed by the following form of the Nernst Equation:

$$E_H = E^0 \pm 2.3 \frac{RT}{nF} \log \left(\frac{Ox}{Red} \right)$$

where E_H is the potential developed between the metallic electrode and the SHE; E^o is the standard reduction potential; R is the gas constant; T is the temperature in ${}^{\circ}K$; n is the number of electrons involved in the reaction; F is Faraday's constant; Ox is the activity of the oxidized form and Red the activity of the reduced form. Since the SHE is rarely used in practice, the measured potential (E_{meas}) will not be equal to E_H , and E_{meas} must be converted to E_H , using the equation:

$$E_H = E_{meas} + E_{ref}$$

where E_{ref} is the reduction potential of the reference electrode.

ORP Measurement Systems

Oxidation-Reduction Potential (ORP) measurements are displayed in mV, using a metallic indicating electrode and standard reference electrode. In principle, ORP measurements should not require standardization; in practice, it may be necessary to check the system against standards of known potential, as described in ASTM Method D 1498.

The ORP Meter. The meter is usually a pH meter operating in mV mode, and is chosen based on the required resolution: ±5mV, ±1mV, or ±0.1mV. Since automatic temperature compensation does not, in most pH meters, operate in mV mode, and since ORP measurements do vary with temperature, the meter should be adjusted to read 0mV with inputs shorted, for readout in absolute mV. For relative mV readout, the meter is set to some arbitrary value with inputs shorted, or with electrodes immersed in a standard solution.

The Electrode System. This consists of two half cells: a metallic indicating electrode—usually of platinum, gold, silver or mercury—to measure the potential of the reaction, and a reference electrode to provide a constant potential and complete the electrical circuit. The reference, as with pH measurements, should be chosen for the compatibility of its junction type and electrolyte with the sample. (See p. 15 for a discussion of reference electrode selection. See p. 22 for Accumet metallic electrodes.)

ISE Theory and Measurement

ISE Theory

Ion-Selective Electrodes (ISEs) respond to a particular ion in solution because of the characteristics of the electrode's sensing membrane. Ideally, the ISE develops an electrical potential which is proportional to the concentration of the ion for which the membrane is selective. The most widely-used ISE is the glass-membrane pH electrode; its use is covered on pp.14–15. This section discusses some non-pH types of ISEs.

When an ISE—the indicator electrode—and a reference electrode are placed in a solution and connected to a pH/ion meter, they form a potentiometric cell. At equilibrium, the meter measures the potential differ-

ence between the ISE and the reference electrode. This potential is proportional to the activity of the ion of interest, and the relationship is described by the Nernst Equation:

$$E = E^0 \pm 2.3 \frac{RT}{nE}$$
 (log A)

where E is the measured electrode potential; E^0 is the standard potential of the system; R is the gas constant; T is the temperature in ${}^{\circ}K$; F is Faraday's constant; A is the activity of the ion being measured; and n is the number of electrons involved in the reaction.

Activity [A] is not the same as concentration. The activity of an ion is strongly influenced by the total ionic strength of the solution. When ISE measurements are made, a determination in concentration units is usually desired. This is done by adding an Ionic Strength Adjuster (ISA) to samples and standards. When ionic strength is held constant, the Nernst Equation reduces to:

$$E = E^0 \pm 2.3 \frac{RT}{nF} \text{ (log C)}$$

where C is the concentration of the ion of interest. This form of the Nernst Equation states that the electrode potential varies directly with the log of the concentration, in a straight-line manner. The slope of the line is equal to the value of:

The table below gives theoretical slope values at 25°C :

Species	Slope (mV/decade)
Monovalent cation	+59.16
Monovalent anion	-59.16
Divalent cation	+29.58
Divalent anion	-29.58

The equation is as follows:

$$S = 2.3 \frac{RT}{nF}$$

where ${\bf S}$ is the electrode's slope. The Nernst Equation thus reduces to:

$$E = E^0 \pm S (log C)$$

ISE Analysis Methods

Direct Analysis. Although this method is called "direct," some sample preparation is usually needed. Normally, an ionic strength adjuster and/or pH adjuster must be added to samples and standards. Then, standards are used to calibrate a meter, or to construct a calibration curve (by plotting the electrode's output in mV vs. the log of concentration). Sample concentration is then read directly from the meter or calibration curve.

Incremental Methods. These procedures can reduce errors caused by temperature variations, complex matrices and complexation. They're also useful for applications where only occasional samples are analyzed.

Incremental methods include: 1) Known Addition; 2) Known Subtraction; 3) Analate Addition; and 4) Analate Subtraction.

With the **Known** methods, the initial electrode output is measured in the sample solution; then an aliquot of standard is added, and the final electrode output is measured. The concentration of the sample can then be calculated from the change in the electrode's output. Equations for known addition and subtraction follow. Known Addition:

$$C_{\text{S}} = \frac{C_{\text{std}} \; V_{\text{std}}}{\left(V_{\text{S}} + V_{\text{std}}\right) \; 10^{\Delta E/S} - V_{\text{S}}} \label{eq:cs}$$

Known Subtraction:

$$C_s = \frac{C_{std} V_{std}}{V_{s-1}(V_{s+1} + V_{std}) 10^{\Delta E/S}}$$

where C_s is the concentration of the sample; C_{std} is the concentration of the standard; V_s is the sample volume; V_{std} is the volume of standard; ΔE is the change in electrode output; and S is the electrode's slope.

In the Analate methods, the initial reading is taken with the electrode immersed in the standard; then an aliquot of sample is added, and the final reading taken. Sample concentration is then calculated from the change in electrode output.

Analate Addition:

$$C_s = C_{std} \left[\left(\frac{V_s + V_{std}}{V_s} \right) 10^{\Delta E/S} - \left(\frac{V_{std}}{V_s} \right) \right]$$

Analate Subtraction:

$$C_s = C_{std} \, \left[\left(\frac{V_{std}}{V_s} \right) - \left(\frac{V_s + V_{std}}{V_s} \right) \, 10^{\Delta E/S} \right]$$

With modern ISE meters, operating in Concentration, Known Addition or Analate Addition modes, the units of the sample result will be identical to those of the standards used in calibration. However, when operating in either the Known Subtraction or Analate Subtraction modes, the ion of interest in the sample is not identical to the standard species. So it's necessary to enter the standard values in concentration units which take into detailed account the chemical relationship between the standard species and the ion of interest in the sample. The proper choice may be referred to as "stoichiometric equivalency units."

Titrations. Ion-selective electrodes can also be used to detect the endpoint of a titration. The ISE can be selected to monitor either the addition of titrant or the depletion of analate. The electrode potential is plotted vs. the volume of titrant added. The volume corresponding to the equivalence point is determined from the graph, and used to calculate sample concentration.

A number of metallic electrodes are also used in titrations. Dual Platinum Wire and Plate Electrodes, are used with pH meters and titration instruments in dead-

stop and amperometric titrations; and Silver Billet Electrodes are the choice for silver and halide titrations. (Accumet® metallic electrode specifications are found on p. 22.)

ISE Measurement Systems

Potentiometric systems for the measurement of a specific ion include an ion meter, calibration standards, and an electrode system consisting of an indicating half cell and a reference half cell.

The Ion Meter. In general, the meter should have a resolution of at least 1mV; a meter with 0.1mV resolution or better is preferred. For direct-read measurements, a meter with concentration mode is necessary; newer microprocessor-based models provide resolution to as many as three significant digits, with excellent accuracy and reproducibility. And for incremental methods, meters now available that will automatically perform the needed calculations, providing readout in ppm, %, mg/mL, mg/L, or most any units you choose. Some current meters have the ability to measure via multiple inputs, permitting simultaneous testing of different samples or measurement of more than one ion in a single sample.

ISE Standardization and Adjustment Solutions. The ion meter/electrode system must be standardized by immersing the electrodes into solutions having a known concentration of the ion of interest. ISE standards are available in a variety of molar, ppm and percent concentrations. Calibration may be done at a single point, usually on less sophisticated meters; or at up to five standardization points, using newer meters.

Additionally, an ionic strength adjuster is required to eliminate interference from other ions and permit readout in units of concentration rather than activity. Since some ISEs have a restricted pH range, a pH adjustment solution may also be necessary. Other special reagents and solutions are available for specific applications.

The Ion-Selective Electrode System. The electrode system consists of: an indicating electrode, which develops a potential dependent on the ionic activity of the sample; and a reference electrode, which provides a constant potential and completes the circuit. The indicating electrode is chosen for its specific response in solutions containing the ion to be measured. The reference electrode is selected based on sample compatibility. Some ISEs are available as combination electrodes, with the indicating half-cell and reference half-cell joined coaxially in a single body.

(See p. 15 for a discussion of reference electrode selection.) (See pp. 19 and 23 for Accumet ISEs.)

Testing and Standardization of pH Meters and Electrodes

Testing the pH meter:

- For most current meters, attach the BNC shorting cap to the meter's BNC input. (For older meters, see your instruction manual.)
- Place the meter in mV mode. Meter's display should now read 0.0mV, ±0.2mV. If the reading is outside of this range, the meter may need to be electronically calibrated.

Testing the electrodes:

- Place the meter in mV mode and measure mV in pH 4 and 7 buffers.
- Determine the net change in millivolts.
 Example: if the pH 7 buffer reading was
 -10.0mV, and the pH 4 buffer reading was
 159.1mV, the net change would be 169.1mV.

- Divide the result (net change) by 177.5, then multiply by 100 to determine the % of the electrode's slope. (Example: 169.1/177.5 x 100 = 95.3%.)
- New electrodes—fresh out of the box—have a slope between 95.0% and 102%. If the slope drops below 92%, clean the electrode.
- If the slope remains below 90.0% or above 102% after cleaning, replace the electrode.

Standardization of electrodes:

- Standardization should be performed at least once a day, and preferably every 2 hours.
- The filling hole should remain open, unless the electrode will be placed in long-term storage.
- At least 2 buffers should be used—one below the expected pH of the sample, and one above the sample

- pH—to provide linear response in the area of interest.
- Use fresh buffers, stir, then immerse electrode (and an ATC probe) into first buffer. Standardize according to the meter's instructions.
- Rinse the electrode, and place it into second buffer.
 Once the meter has accepted the second buffer, it will calculate the electrode's % slope.
- New electrodes—fresh out of the box—have a slope between 95.0% and 102%. If the slope drops below 92%, clean the electrode.

Basic pH Electrode Care and Troubleshooting Tips

Liquid-Filled Calomel Electrodes and Ag/AgCl Double-Junction Electrodes:

Start up:

- Remove the end cot and open the fill hole.
 Add filling solution if necessary.
- Soak at least 2 hours, and preferably overnight, in pH 4 buffer solution.

Storage for less than 1 week:

- Soak the electrode in SE40-1 electrode storage solution, pH 4 or pH 7 buffer solution, with fill hole OPEN.
- Never store electrodes in water; after rinsing, blot them dry—never wipe.

Storage for longer than 1 week:

Saturate the cotton in the end cot with pH
 4 buffer and replace the cot on the glass
 bulb of the electrode. Close the fill hole
 and pack the electrode in the original box.

Unblocking the reference junction:

- Soak the electrode tip in 60°C distilled water for 5 to 10 minutes.
- Check electrode for flow by first blotting the reference junction dry with a wiper (e.g., Kimwipe[®]). Pressurize the electrode

for 10 seconds, then blot the reference junction again. If wiper picks up liquid, electrode is flowing.

Liquid-Filled Ag/AgCl Single-Junction Electrodes:

Start up

- Remove the end cot and open the fill hole. Add filling solution if necessary.
- Soak at least 2 hours, and preferably overnight, in pH 4 buffer solution.

Storage for less than 1 week:

- Soak the electrode in SE40-1 electrode storage solution, pH 4 or pH 7 buffer solution, with fill hole OPEN.
- Never store electrodes in water; after rinsing, blot them dry—never wipe.

Storage for longer than 1 week:

 Saturate the cotton in the end cot with pH 4 buffer and replace the cot on the glass bulb of the electrode. Close the fill hole and pack the electrode in the original box.

Unblocking the reference junction:

- Soak the electrode tip in 60°C saturated potassium chloride for 5 to 10 minutes.
- Check electrode for flow by first blotting the reference junction dry with a wiper

(e.g., Kimwipe). Pressurize the electrode for 10 seconds, then blot the reference junction again. If wiper picks up liquid, electrode is flowing.

Gel-Filled Ag/AgCl Single- or Double-Junction Electrodes:

Start up:

- Remove the end cot from the electrode's tip.
- Soak at least 2 hours, and preferably overnight, in pH 4 buffer solution.

Storage for less than 1 week:

 Soak the electrode in SE40-1 electrode storage solution, pH 4 or pH 7 buffer solution.

Storage for longer than 1 week:

- Saturate the cotton in the end cot with pH 4 buffer and replace the cot on the glass bulb of the electrode. Pack the electrode in the original box, and store in upright position.
- Never store electrodes in water; after rinsing, blot them dry—never wipe.

Unblocking the reference junction:

 Soak the electrode tip in 60°C saturated potassium chloride solution for 5 to 10 minutes.

Accumet® Electrodes

for pH, Titrations, ORP, and Ion Selective Electrode Analysis

- Over 30 years of experience in the design, development, and manufacture of electrodes go into each Accumet electrode from Fisher Scientific.
- Each electrode is individually tested, serialized to meet GLP requirements, and backed by Fisher's Technical Applications staff.
- No matter what the application, we offer an ideal electrode for fast, accurate pH, titration, ORP, conductivity or ISE measurements.

Accumet® pH Electrodes:

 AccuFlow™ Flushable Junction pH/ATC and pH Electrodes



Perfect choice for even the most difficult samples! Easy-clean junction, plastic body, choice of AccuopHast and Ag/AgCl reference types, plus models with ATC element built in for small samples. (See specifications on p. 21.)

• AccuTupH® Rugged Bulb pH Electrodes



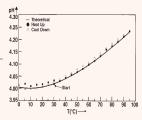
Best choice for durability! Made to be up to 40 times tougher than conventional glass pH electrodes, without sacrificing response time. Available in pH half cell, variable-temperature combination and standard combination models. (See specifications on p. 21.)

AccuFet® Solid-State pH/ATC Electrodes



Solid state pH/temperature electrodes—the perfect choice for situations where breakable glass electrodes are hazardous. (See specifications on p. 22.)

Accu • pHast[®] Variable
 Temperature Electrodes



Fast, accurate pH measurements! Glass- and epoxy-bodies, patented reference design. Deliver superlative performance even in samples with widely varying temperatures.

(See specifications on p. 21.)

 Accumet pH, Reference and Combination Electrodes



Wide choice: glass or plastic, calomel or Ag/AgCl, low maintenance, single- or double- junction, liquid or gel filled, flat surface and microsize electrodes.

(See specifications on pp. 21-23.)

All Accumet refillable electrodes feature our patented, easy-to-use fill hole technology.



• Accumet Conductivity Cells:



Two-cell models (0.1, 1.0 and 10.0cm⁻¹ cell constants). Plus new 4 cell models with cell constants of 1.0 and 10.0cm⁻¹. Glass or epoxy bodies.

(See specifications on p. 23.)

Accumet Metallic Electrodes:



We offer five models, for ORP, silver/halide, Karl Fischer and chlorine measurements.

(See specifications on p. 22.)

Accumet ISE Electrodes:

 Accumet Polymer Membrane Combination ISEs



Calcium, Nitrate and Potassium Combinations feature rugged epoxy bodies and refillable reference section. (See specifications on p. 23.)

Accumet Solid-State ISEs



Cyanide Combination has durable epoxy body, homogeneous membrane. Bromide, Chloride, Copper, Fluoride, Lead and Silver/Sulfide half cells have epoxy bodies, sensing membrane with low microporosity and mirror-like finish to minimize sample retention. Bromide, Chloride, Copper, Fluoride, Lead and Silver/Sulfide Combinations have glass bodies, excellent range and precision. (See specifications on p. 23.)

• Accumet Gas-Sensing Combination ISEs



Ammonia and Carbon Dioxide models feature simple two-piece construction; tough, durable body that's highly resistant to chemicals; permanently-bonded membrane/cap unit. (See specifications on p. 23.)

Accumet Glass-Membrane ISEs



Sodium Glass-Body Half Cell and Sodium Glass-Body Combination have specially-formulated sodium-responsive glass bulb for direct measurement of active sodium-ion concentration. (See specifications on p. 23.)

Accumet ISE/pH Reference Electrodes



Single-Junction and Double-Junction types are excellent for ISE determinations, provide premium performance in pH and titration applications, too.

(See specifications on p. 23.)

Select the Right pH Electrode for Your Application

	<u> </u>				
Application/ Sample Type His	Accu • pHast® ghest Speed and Accuracy	Accumet® Ag/AgCl Standard Line	Accumet Calomel (Tris Compatible)	Accumet Gel-Filled Line	AccuFet® Nonglass pH
Please NOTE : the numbers liste	ed below are partial catalog	numbers13-620-X.			
General Purpose	-296 ^B , -281 ^A	-285 ^B , -90 ^A	-286 ^B , -271 ^A	-108 ^B , -104 ^A	
Most Sample Types	-298 ^B , -279 ^A	-223 ^B , -222 ^A , -110 ^{B/M} , -109 ^B		-299 ^B , -283 ^A	
	-185 ^B , -184 ^A	-284 ^B /-46 ^P , -3 ^A /-46 ^P			
	-116 ^{B/M} , -117 ^B	-294 ^B /-46 ^P , -256 ^A /-46 ^P			
Biological/	-296 ^B , -281 ^A , -116 ^{B/M}	-183 ^B , -182 ^A	-286 ^B , -271 ^A	-299 ^B , -283 ^A	
Pharmaceutical	-297 ^B , -280 ^A , -117 ^B	-223 ^B , -222 ^A , -110 ^{B/M} , -109 ^B	-293 ⁸ , -270 ^A		
Proteins, Tris, Enzymes	-185 ^B , -184 ^A	-284 ^B /-52 ^P , -3 ^A /-52 ^P			
Education/	-298 ^B , -279 ^A	-183 ^B , -182 ^A	-300 ^B	-108 ^B , -104 ^A	
Student Use	-113 ^{B/Q} , -114 ^{B/M}	-221 ^B , -220 ^A	-288 ⁸ , -272 ^A	-299 ^B , -283 ^A	
	-185 ^B , -184 ^A	-287 ^B , -97 ^A , -98 ^{P/P}			
Emulsions	-116 ^{B/M}	-110 ^{B/M}	-284 ^B /-62 ^P , -3 ^A /-62 ^P		
Food, Cosmetics, Oils	-11 7 [₿]	-109 ^B	-284 ^B /-61 ^P , -3 ^A /-61 ^P		
Flat Surface		-289 ^B , -83 ^A			
Food, Cheese, Agar		,			
Harsh Environments:					
• Field or Plant Use	-298 ^B , -279 ^A	-287 ^B , -97 ^A , -98 ^{P/P}	-531 ^{B/M} , -AP51 ^{B/W}	-108 ^B , -104 ^A	
	-113 ^{B/Q} , -114 ^{B/M}	-530 ^{B/M} , -AP50 ^{B/W}	-300 ^B	-299 ^B , -283 ^A	
	-116 ^{B/M} , -117 ^B	-AP1 ^T , -110 ^{B/M} , -109 ^B	-288 ^B , -272 ^A	,	
Rugged Use	-185 ⁸ , -184 ^A	-183 ^B , -182 ^A	200 , 272	-108 ^B , -104 ^A	
Roggea Osc	100 , 104	-181 ^B , -180 ^A		-299 ^B , -283 ^A	
		-221 ^B , -220 ^A		-AP52 ^{B/W} , -AP2 ^T	
		-187 ⁸ /-53 ⁹ , -186 ^A /-53 ⁹	-187 ^B /-258 ^P , -186 ^A /-258 ^P	-111 ^{B/M} , -112 ^{B/Q}	
Large Sample Size	-298 ^B , -279 ^A	107 / 33 , 100 / 33	-293 ^B , -270 ^A	-290 ^B , -252 ^A	
Tall Flasks or Bottles				-270 , -232	
Low Ionic Strength	-296 ^B , -281 ^A	-284 ^B /-46 ^P , -3 ^A /-46 ^P	-286 ^B , -271 ^A		
Treated Effluent	-116 ^{B/M} , -117 ^B	-223 ^B , -222 ^A , -110 ^{B/M} , -109 ^B	-284 ^B /-61 ^P , -3 ^A /-61 ^P		
Nonaqueous	-296 ^B , -281 ^A	-284 ^B /-46 ^P , -3 ^A /-46 ^P	-286 ^B , -271 ^A		
Solvents, Alcohols		-223 ^B , -222 ^A	-284 ^B /-57 ^P , -3 ^A /-57 ^P		
Process Monitoring	-185 ^B , -184 ^A	-183 ^B , -182 ^A		-299 ^B , -283 ^A	
		-181 ^B , -180 ^A		-111 ^{B/M} , -112 ^{B/Q}	
Pulp & Paper	-185 ^B , -184 ^A	-183 ^B , -182 ^A			
		-223 ^B , -222 ^A		-299 ^B , -283 ^A	
Semisolids					-755 ^D (all AR/AB)
Fruit, Meat, Cheese					-758 ^{MD} + -759 ^B
					-AP20 [™]
Small Sample Size	-297 ^B , -280 ^A	-94 ^A	-293 ^B , -270 ^A	-290 ^B , -252 ^A	
Test Tubes, Small Flasks		-291 ^B , -92 ^A			
		-292 ^B , -93 ^A			
Titration	-296 ^B , -281 ^A	-223 ^B , -222 ^A	-286 ^B , -271 ^A		
		-183 ^B , -182 ^A			
		-181 ^B , -180 ^A			
Viscous Samples	-116 ^{B/M}	-110 ^{B/M}	-284 ^B /-61 ^P , -3 ^A /-61 ^P		
Slurries, Sludges	-11 7 [₿]	-109 ^B			
Waters:					
Acid Rain, Boiler Fe Distillad Built Wall		-223 ^B , -222 ^A	-284 ^B /-61 ^P , -3 ^A /-61 ^P		
Distilled, Rain, Well		000R 000A	-286 ⁸ , -271 ^A	100R 104A	
 Drinking, Tap 	-296 ⁸ , -281 ^A	-223 ^B , -222 ^A	-284 ^B /-61 ^P , -3 ^A /-61 ^P	-108 ^B , -104 ^A	
	-185 ^B , -184 ^A	-284 ^B /-46 ^P , -3 ^A /-46 ^P	-286 ^B , -271 ^A	-299 ^B , -283 ^A	
• Sea Water	-296 ^B , -281 ^A	-223 ^B , -222 ^A			
	-185 ^B , -184 ^A				
 Wastewater 	-296 ^B , -281 ^A	-284 ^B /-46 ^P , -3 ^A /-46 ^P		-108 ^B , -104 ^A	
	-185 ^B , -184 ^A	-223 ^B , -222 ^A		-299 ^B , -283 ^A	

^{*}With U.S. Standard connector (and Pin connector on combinations). *With BNC connector. ** With BNC and Mini-Phone ATC connectors. ** With BNC and Mini-Phone ATC connectors. ** With DIN connector. ** With Pin connector. ** With Pin connector. ** With Pin connector. ** With Twist-Lock waterproof connector. ** With Pin co

^{/-} Indicating Electrode/Reference Electrode

Catalog No.	Description	pH Range	Temp. Range (°C)	Length (mm)	Diameter (mm)	Typical Applications
	ushable Junction, Plastic Bod			•		
Vith Accu• pH as		,	,			
3-620-117 ^{B/M}	pH/ATC	0–14	0–100	175	12	General Purpose, Tris,
13-620-116 ⁸	pH	0–14	0–100	175	12	soils, sludges, viscous
With Double Jun 13-620-110™	ction Reterence pH/ATC	0–14	0–100	175	12	General Purpose, Tris,
13-620-110°	рН	0-14	0–100 0–100	175	12	soils, sludges, viscous
	Variable Temperature Comb			170	12	Jone, Sidagos, Viscodo
13-620-281^	Standard Glass Body	0–14	-5 - 100	102	10	General Purpose,
13-620-296 ⁸						Tris, Water
13-620-280 ^a	MicroProbe™ Glass	0-14	-5–100	165	5	Small Samples,
13-620-297 ^B	Body					Tris
13-620-279	Extra-Long Epoxy	0–14	-5–100	140	10	Field/Plant Use,
13-620-298 ⁸ 13-620-113 ^{8/0}	Body pH/ATC, Extra-Long	0-14	-5–100	143	10	Tris Field/Plant Use,
10-020-110	Epoxy Body	0-14	-3-100	140	10	Tris
13-620-114 ^{B/M}	pH/ATC, Extra-Long	0–14	-5-100	143	10	Field/Plant Use,
	Epoxy Body					Tris
AccuTupH+®	Variable Temperature Rugge	d Bulb Comb	bination pH Electro	odes •		
13-620-184	Double Junction	0–14	-5-100	102	10	General Purpose,
13-620-185 ⁸						Tris, Rugged
	uTupH® Rugged Bulb Glass					
13-620-180^	Single Junction,	0–14	0–100	102	10	General Purpose,
3-620-181 ⁸ 3-620-182 ^A	Ag/AgCI Reference Double Junction,	0–14	0–100	102	10	Rugged Use General Purpose,
3-620-162"	Ag/AgCl Reference	0-14	0-100	102	10	Tris, Water
	dard Size Glass Body Comb	vination nH F	-lectrodes •			ino, wator
13-620-90 ^A	Single Junction,	0–14	-5–100	106	10	General Purpose
13-620-285 ⁸	Ag/AgCl Reference					
13-620-222 ^a	Double Junction,	0-14	-5-100	106	10	General Purpose,
13-620-223 ^B	Ag/AgCl Reference					Tris, Titration
13-620-271	Single Junction, Calomel Reference	0–14	-5–80	106	10	General Purpose, Tris, Water
13-620-286 ⁸						ITIS, Water
3-620-94 ^A	oProbe™ Glass Body Combine Micro tip, Miniature Length,	nation pH Eli 0–14	-5–110	60	6	Small Samples
3-020-94	Ag/AgCl Reference	0-14	-3-110	00	U	Sman Samples
13-620-92 ^A	Micro tip, Standard Length,	0–14	-5–110	100	6	Small Samples,
3-620-291 ⁸	Ag/AgCl Reference					Test Tubes
3-620-93 ^A	Micro tip, Extra Long,	0-14	-5–110	150	6	Small Samples,
3-620-292 ⁸	Ag/AgCl Reference					Flasks, Tubes
13-620-270^	Semimicro, Extra Long,	0–14	-5–80	150	6	Small Samples,
13-620-293 ⁸	Calomel Reference	. 5:11 0	1			Tris, Test Tubes
Accumet Stand 13-620-530 ^{B/M}	dard Size Polymer Body Liqu		mbination pH Elec -5–100	trodes 106	10	Field/Plant Use
	pH/ATC, Polypropylene	0–14	- 1- 11111	11111	1111	
1.5-NZII-APT	Rody Ad/Adl:1 Reterence		0 100	100	10	rieiu/riaiit USE
	Body, Ag/AgCl Reference pH/ATC, Polypropylene	0–14	-5-80	106	10	
13-620-531 ^в ™	Body, Ag/AgCl Reference pH/ATC, Polypropylene Body, Calomel Reference	0–14		106	10	Field/Plant Use, Tris
13-620-531 ^{8/M}	pH/ATC, Polypropylene Body, Calomel Reference Polypropylene Body,	0–14 0–14				Field/Plant Use, Tris Field/Plant Use,
13-620-531 ^{BM} 13-620-97 ^A 13-620-287 ^B	pH/ATC, Polypropylene Body, Calomel Reference Polypropylene Body, Ag/AgCl Reference	0–14	-5–80 -5–100	106 106	10	Field/Plant Use, Tris Field/Plant Use, Student Use
13-620-531 ^{BM} 13-620-97 ^A 13-620-287 ^B	pH/ATC, Polypropylene Body, Calomel Reference Polypropylene Body, Ag/AgCl Reference Polypropylene Body,		-5-80	106	10	Field/Plant Use, Tris Field/Plant Use, Student Use Field/Plant Use,
13-620-531 ^{B/M} 13-620-97 ^A 13-620-287 ^B 13-620-300 ^B	pH/ATC, Polypropylene Body, Calomel Reference Polypropylene Body, Ag/AgCl Reference Polypropylene Body, Calomel Reference	0–14 0–14	-5–80 -5–100 -5–80	106 106 106	10 10 10	Field/Plant Use, Tris Field/Plant Use, Student Use Field/Plant Use, Tris, Student Use
3-620-531 ^{BM} 13-620-97 ^A 13-620-287 ^B 13-620-300 ^B 13-620-220 ^A	pH/ATC, Polypropylene Body, Calomel Reference Polypropylene Body, Ag/AgCl Reference Polypropylene Body, Calomel Reference Polypropylene Body,	0–14	-5–80 -5–100	106 106	10	Field/Plant Use, Tris Field/Plant Use, Student Use Field/Plant Use, Tris, Student Use Rugged Use,
13-620-531 ^{B,M} 13-620-97 ^A 13-620-287 ^B 13-620-300 ^B 13-620-220 ^A 13-620-221 ^B	pH/ATC, Polypropylene Body, Calomel Reference Polypropylene Body, Ag/AgCl Reference Polypropylene Body, Calomel Reference Polypropylene Body, Double Junction, Ag/AgCl	0–14 0–14	-5–80 -5–100 -5–80	106 106 106	10 10 10 10	Field/Plant Use, Tris Field/Plant Use, Student Use Field/Plant Use, Tris, Student Use
13-620-531 ^{BM} 13-620-97 ^A 13-620-287 ^B 13-620-300 ^B 13-620-220 ^A 13-620-221 ^B 13-620-272 ^A	pH/ATC, Polypropylene Body, Calomel Reference Polypropylene Body, Ag/AgCl Reference Polypropylene Body, Calomel Reference Polypropylene Body,	0–14 0–14 0–14	-5–80 -5–100 -5–80 -5–100	106 106 106 106	10 10 10	Field/Plant Use, Tris Field/Plant Use, Student Use Field/Plant Use, Tris, Student Use Rugged Use, Tris, Student Use
3-620-531 ^{B,M} 3-620-97 ^A 3-620-287 ^B 3-620-300 ^B 3-620-220 ^A 3-620-221 ^B 3-620-272 ^A 3-620-288 ^B 3-620-83 ^A	pH/ATC, Polypropylene Body, Calomel Reference Polypropylene Body, Ag/AgCl Reference Polypropylene Body, Calomel Reference Polypropylene Body, Double Junction, Ag/AgCl Epoxy Body, Calomel Reference Epoxy Body, Flat Surface,	0–14 0–14 0–14	-5–80 -5–100 -5–80 -5–100	106 106 106 106	10 10 10 10	Field/Plant Use, Tris Field/Plant Use, Student Use Field/Plant Use, Tris, Student Use Rugged Use, Tris, Student Use Field/Plant Use, Tris, Student Use Field/Plant Use, Tris, Student Use
3-620-531 ^{B,M} 3-620-97 ^A 3-620-287 ^B 3-620-300 ^B 3-620-220 ^A 3-620-221 ^B 3-620-272 ^A 3-620-288 ^B 3-620-83 ^A 3-620-289 ^B	pH/ATC, Polypropylene Body, Calomel Reference Polypropylene Body, Ag/AgCl Reference Polypropylene Body, Calomel Reference Polypropylene Body, Double Junction, Ag/AgCl Epoxy Body, Calomel Reference Epoxy Body, Flat Surface, Ag/AgCl Reference	0–14 0–14 0–14 0–14 0–14	-5–80 -5–100 -5–80 -5–100 -5–80 -5–100	106 106 106 106 106	10 10 10 10 10	Field/Plant Use, Tris Field/Plant Use, Student Use Field/Plant Use, Tris, Student Use Rugged Use, Tris, Student Use Field/Plant Use, Tris, Student Use
3-620-531 ^{B,M} 3-620-97 ^A 3-620-287 ^B 3-620-300 ^B 3-620-221 ^B 3-620-272 ^A 3-620-288 ^B 3-620-289 ^B Accumet Polyr	pH/ATC, Polypropylene Body, Calomel Reference Polypropylene Body, Ag/AgCl Reference Polypropylene Body, Calomel Reference Polypropylene Body, Double Junction, Ag/AgCl Epoxy Body, Calomel Reference Epoxy Body, Flat Surface, Ag/AgCl Reference mer Body Gel-Filled Combin	0–14 0–14 0–14 0–14 0–14 ation pH Ele	-5-80 -5-100 -5-80 -5-100 -5-80 -5-100	106 106 106 106 106 114	10 10 10 10 10 10	Field/Plant Use, Tris Field/Plant Use, Student Use Field/Plant Use, Tris, Student Use Rugged Use, Tris, Student Use Field/Plant Use Field/Plant Use, Tris, Student Use Food, Cheese, Agar, Paper
13-620-104^	pH/ATC, Polypropylene Body, Calomel Reference Polypropylene Body, Ag/AgCl Reference Polypropylene Body, Calomel Reference Polypropylene Body, Double Junction, Ag/AgCl Epoxy Body, Calomel Reference Epoxy Body, Flat Surface, Ag/AgCl Reference mer Body Gel-Filled Combin Polypropylene Body, gelled	0–14 0–14 0–14 0–14 0–14	-5–80 -5–100 -5–80 -5–100 -5–80 -5–100	106 106 106 106 106	10 10 10 10 10	Field/Plant Use, Tris Field/Plant Use, Student Use Field/Plant Use, Tris, Student Use Rugged Use, Tris, Student Use Field/Plant Use Field/Plant Use, Tris, Student Use Food, Cheese, Agar, Paper
13-620-531 ^{B,M} 13-620-97 ^A 13-620-287 ^B 13-620-300 ^B 13-620-220 ^A 13-620-221 ^B 13-620-272 ^A 13-620-288 ^B 13-620-289 ^B Accumet Polyr 13-620-104 ^A 13-620-108 ^B	pH/ATC, Polypropylene Body, Calomel Reference Polypropylene Body, Ag/AgCl Reference Polypropylene Body, Calomel Reference Polypropylene Body, Double Junction, Ag/AgCl Epoxy Body, Calomel Reference Epoxy Body, Flat Surface, Ag/AgCl Reference mer Body Gel-Filled Combin Polypropylene Body, gelled Ag/AgCl Reference	0-14 0-14 0-14 0-14 0-14 ation pH Electors	-5-80 -5-100 -5-80 -5-100 -5-80 -5-100	106 106 106 106 106 114	10 10 10 10 10 10 13	Field/Plant Use, Tris Field/Plant Use, Student Use Field/Plant Use, Tris, Student Use Rugged Use, Tris, Student Use Field/Plant Use, Tris, Student Use Field/Plant Use, Agar, Paper General Purpose, Field/Plant Use
13-620-531 ^{B,M} 13-620-97 ^A 13-620-287 ^B 13-620-300 ^B 13-620-220 ^A 13-620-221 ^B 13-620-272 ^A 13-620-288 ^B 13-620-289 ^B Accumet Polyr 13-620-104 ^A 13-620-108 ^B 13-620-252 ^A	pH/ATC, Polypropylene Body, Calomel Reference Polypropylene Body, Ag/AgCl Reference Polypropylene Body, Calomel Reference Polypropylene Body, Double Junction, Ag/AgCl Epoxy Body, Calomel Reference Epoxy Body, Flat Surface, Ag/AgCl Reference mer Body Gel-Filled Combin Polypropylene Body, gelled Ag/AgCl Reference Epoxy Body, Pencil-Thin,	0–14 0–14 0–14 0–14 0–14 ation pH Ele	-5-80 -5-100 -5-80 -5-100 -5-80 -5-100	106 106 106 106 106 114	10 10 10 10 10 10	Field/Plant Use, Tris Field/Plant Use, Student Use Field/Plant Use, Tris, Student Use Rugged Use, Tris, Student Use Field/Plant Use Field/Plant Use, Tris, Student Use Food, Cheese, Agar, Paper
13-620-531 ^{B,M} 13-620-97 ^A 13-620-287 ^B 13-620-300 ^B 13-620-221 ^B 13-620-272 ^A 13-620-288 ^B 13-620-289 ^B Accumet Polyr 13-620-104 ^A 13-620-108 ^B 13-620-252 ^A 13-620-290 ^B	pH/ATC, Polypropylene Body, Calomel Reference Polypropylene Body, Ag/AgCl Reference Polypropylene Body, Calomel Reference Polypropylene Body, Double Junction, Ag/AgCl Epoxy Body, Calomel Reference Epoxy Body, Flat Surface, Ag/AgCl Reference mer Body Gel-Filled Combin Polypropylene Body, gelled Ag/AgCl Reference Epoxy Body, Pencil-Thin, gelled Ag/AgCl Reference	0-14 0-14 0-14 0-14 0-14 ation pH Electors	-5-80 -5-100 -5-80 -5-100 -5-80 -5-100	106 106 106 106 106 114 106 178	10 10 10 10 10 13	Field/Plant Use, Tris Field/Plant Use, Student Use Field/Plant Use, Tris, Student Use Rugged Use, Tris, Student Use Field/Plant Use, Tris, Student Use Field/Plant Use, Tris, Student Use Food, Cheese, Agar, Paper General Purpose, Field/Plant Use Tall Flasks, Bottles
13-620-531 ^{B,M} 13-620-97 ^A 13-620-287 ^B 13-620-300 ^B 13-620-220 ^A 13-620-221 ^B 13-620-272 ^A 13-620-288 ^B 13-620-289 ^B Accumet Polyr 13-620-104 ^A 13-620-108 ^B 13-620-252 ^A 13-620-290 ^B 13-620-283 ^A	pH/ATC, Polypropylene Body, Calomel Reference Polypropylene Body, Ag/AgCl Reference Polypropylene Body, Calomel Reference Polypropylene Body, Double Junction, Ag/AgCl Epoxy Body, Calomel Reference Epoxy Body, Flat Surface, Ag/AgCl Reference mer Body Gel-Filled Combin Polypropylene Body, gelled Ag/AgCl Reference Epoxy Body, Pencil-Thin,	0-14 0-14 0-14 0-14 0-14 ation pH Ele 0-14 0-14	-5-80 -5-100 -5-80 -5-100 -5-80 -5-100 -5-80 -5-80	106 106 106 106 106 114	10 10 10 10 10 10 13	Field/Plant Use, Tris Field/Plant Use, Student Use Field/Plant Use, Tris, Student Use Rugged Use, Tris, Student Use Field/Plant Use, Tris, Student Use Field/Plant Use, Agar, Paper General Purpose, Field/Plant Use
3-620-531 ^{B,M} 3-620-97 ^A 3-620-287 ^B 3-620-300 ^B 3-620-221 ^B 3-620-221 ^B 3-620-288 ^B 3-620-289 ^B 3-620-289 ^B 3-620-104 ^A 3-620-108 ^B 3-620-252 ^A 3-620-290 ^B 3-620-283 ^A 3-620-299 ^B 3-620-290 ^B 3-620-299 ^B 3-620-290 ^B 3-620-299 ^B 3-620-290 ^B 3-620-200 ^B 3-620-200 ^B 3-620-200 ^B 3-620-200 ^B 3-620-200 ^B 3-620-200 ^B 3-6	pH/ATC, Polypropylene Body, Calomel Reference Polypropylene Body, Ag/AgCl Reference Polypropylene Body, Calomel Reference Polypropylene Body, Double Junction, Ag/AgCl Epoxy Body, Calomel Reference Epoxy Body, Flat Surface, Ag/AgCl Reference mer Body Gel-Filled Combin Polypropylene Body, gelled Ag/AgCl Reference Epoxy Body, Pencil-Thin, gelled Ag/AgCl Reference Polypropylene Body, Double Junction, Ag/AgCl pH/ATC, Polypropylene Body,	0-14 0-14 0-14 0-14 0-14 ation pH Ele 0-14 0-14	-5-80 -5-100 -5-80 -5-100 -5-80 -5-100 -5-80 -5-80	106 106 106 106 106 114 106 178	10 10 10 10 10 13	Field/Plant Use, Tris Field/Plant Use, Student Use Field/Plant Use, Tris, Student Use Rugged Use, Tris, Student Use Field/Plant Use, Tris, Student Use Field/Plant Use, Tris, Student Use Food, Cheese, Agar, Paper General Purpose, Field/Plant Use Tall Flasks, Bottles General Purpose,
13-620-531 ^{B,M} 13-620-97 ^A 13-620-287 ^B 13-620-300 ^B 13-620-220 ^A 13-620-221 ^B 13-620-288 ^B 13-620-288 ^B 13-620-289 ^B Accumet Polyr 13-620-104 ^A 13-620-108 ^B 13-620-252 ^A 13-620-290 ^B 13-620-290 ^B 13-620-299 ^B 13-620-AP2 ^T	pH/ATC, Polypropylene Body, Calomel Reference Polypropylene Body, Ag/AgCl Reference Polypropylene Body, Calomel Reference Polypropylene Body, Double Junction, Ag/AgCl Epoxy Body, Calomel Reference Epoxy Body, Flat Surface, Ag/AgCl Reference Epoxy Body, Flat Surface, Ag/AgCl Reference mer Body Gel-Filled Combin Polypropylene Body, gelled Ag/AgCl Reference Epoxy Body, Pencil-Thin, gelled Ag/AgCl Reference Polypropylene Body, Double Junction, Ag/AgCl pH/ATC, Polypropylene Body, Double Junction, Ag/AgCl	0-14 0-14 0-14 0-14 0-14 0-14 0-14 0-14	-5-80 -5-100 -5-80 -5-100 -5-80 -5-100 ctrodes -5-80 -5-80 -5-80 -5-80	106 106 106 106 106 114 106 178 106	10 10 10 10 10 13 10 6 10	Field/Plant Use, Tris Field/Plant Use, Student Use Field/Plant Use, Tris, Student Use Rugged Use, Tris, Student Use Field/Plant Use, Tris, Student Use Field/Plant Use, Tris, Student Use Food, Cheese, Agar, Paper General Purpose, Field/Plant Use Tall Flasks, Bottles General Purpose, Tris, Paper/Pulp Field Use
3-620-531 ^{B,M} 3-620-97 ^A 3-620-287 ^B 3-620-300 ^B 3-620-221 ^B 3-620-221 ^B 3-620-288 ^B 3-620-289 ^B 3-620-289 ^B 3-620-104 ^A 3-620-108 ^B 3-620-252 ^A 3-620-290 ^B 3-620-283 ^A 3-620-299 ^B 3-620-290 ^B 3-620-299 ^B 3-620-290 ^B 3-620-299 ^B 3-620-290 ^B 3-620-200 ^B 3-620-200 ^B 3-620-200 ^B 3-620-200 ^B 3-620-200 ^B 3-620-200 ^B 3-6	pH/ATC, Polypropylene Body, Calomel Reference Polypropylene Body, Ag/AgCl Reference Polypropylene Body, Calomel Reference Polypropylene Body, Double Junction, Ag/AgCl Epoxy Body, Calomel Reference Epoxy Body, Flat Surface, Ag/AgCl Reference mer Body Gel-Filled Combin Polypropylene Body, gelled Ag/AgCl Reference Epoxy Body, Pencil-Thin, gelled Ag/AgCl Reference Polypropylene Body, Double Junction, Ag/AgCl pH/ATC, Polypropylene Body,	0-14 0-14 0-14 0-14 0-14 ation pH Ele 0-14 0-14 0-14	-5-80 -5-100 -5-80 -5-100 -5-80 -5-100 ctrodes -5-80 -5-80 -5-80	106 106 106 106 106 114 106 178 106	10 10 10 10 10 13 10 6 10	Field/Plant Use, Tris Field/Plant Use, Student Use Field/Plant Use, Tris, Student Use Rugged Use, Tris, Student Use Field/Plant Use, Tris, Student Use Field/Plant Use, Tris, Student Use Food, Cheese, Agar, Paper General Purpose, Field/Plant Use Tall Flasks, Bottles General Purpose, Tris, Paper/Pulp

Catalog No.	Description	pH Range	Temp. Range (°C)	Length (mm)	Diameter (mm)	Typical Applications
	State Combination pH Elec		0.100	1/2	40	110
13-620-755°	pH/ATC Nonglass	0–14	0–100	140	12	Where glass is a concern; for all AR/AB Meters
3-620-758 ^{MD}	pH/ATC Nonglass	0–14	0-100	140	12	Where glass is a concern
13-620-759 ⁸	Adapter 13-620-758 to BNC	0–14	0.400			W/
3-620-AP20 [™]	pH/ATC Nonglass		0-100	140	12	Where glass is a concern
Accumet® <i>pH/A</i> 3-620-AP52 ^{B/W}	ATC Electrodes for AP Serie			100	10	Field Hee
	pH/ATC, Epoxy Body, gelled Ag/AgCl Double Junction	0–14	0–100	102	10	Field Use, AP Series
3-620-AP51 ^{B/W}	pH/ATC, Polypropylene, Calomel, Single Junction	0–14	0–100	102	10	Field Use, AP Series
3-620-AP50 ^{B/W}	pH/ATC, Polypropylene, Ag/AgCl, Single Junction	0–14	0–100	102	10	Field Use, AP Series
Accumet pH-Ind	dicating Half Cell Electrode	es (Require s	eparate reference	half cell)		
13-620-186 ^A	Glass Body, Rugged Bulb	0–14	0-100	102	10	General Purpose,
3-620-187 ^B	pH Half Cell					Rugged Use
3-620-3^ 3-620-284 ⁸	Glass Body, Universal Glass pH Half Cell	0–14	-5–110	102	10	General Purpose, Plant Use
3-620-256 ^A 3-620-294 ^B	Epoxy Body pH Half Cell	0–14	-5–110	102	10	General Purpose, Field/Plant Use
3-620-1	Full Range, High pH	0–14	-5-110	102	10	General Purpose,
3-620-295 ⁸	Glass, Half Cell	0 11	0 110	TOL	10	Field/Plant Use
Accumet Calor	nel Reference Half Cell Elec	trodes (Regi	uire separate pH-li	ndicating half	cell)	
13-620-51°	Standard Glass Body,	0–14	-580	105	10	General Purpose, Tris
13-620-52°	Dri-Pak Standard Glass Body,	0–14	-5-80	105	10	General Purpose, Tris
	Prefilled					
3-620-79° 3-620-57°	Miniature Glass Body	0–14 0–14	-5–80 -5–80	41 102	10 10	Small Samples
13-020-37	Standard Glass Body, Low Flowrate, Cracked Bead Junction	U-14	-5-60	102	10	Nonaqueous
13-620-62°	Glass Body, High Flowrate, Sleeve Junction	0–14	-5-80	102	14	Viscous Samples, Slurries, Sludges
13-620-61°	Glass Body, High Flowrate, Reverse Sleeve Junction	0–14	-5-80	102	14	Viscous Samples, Slurries, Sludges
13-620-258°	Liquid-Filled, Epoxy Body	0–14	-5-80	102	10	Field/Plant Use
3-620-259°	Gel-Filled, Epoxy Body	0-14	-5-80	102	10	Field/Plant Use
Accumet Ag/A	gCl Reference Half Cell Ele	ectrodes (Red	quire separate pH-	Indicating ha	lf cell) •——	
13-620-53°	Glass Body, Single Junction	0–14	-5–110	105	10	General Purpose
3-620-273°	Glass Body, Double Junction	0–14	-5–110	102	10	General Purpose, Tris
13-620-46 ^p	Single Annular Junction, Epoxy Body, Ag/AgCl	0–14	0–100	108	13	Calcium cyanide, dival- ent cation, fluoride, sodium, redox, pH
13-620-47°	Double Junction (Annular/ Ceramic), Epoxy Body, Ag/AgCl inner, outer cham- ber empty for use with sample compatible electrolyte	0–14	0–100	108	13	Bromide, chloride, cop- per, iodide, lead, nitrate, silver/sulfide; redox and pH applications requiring compatible electrolyte
13-620-45°	Glass Body, Side Arm for Remote Filling	0–14	-5–110	105	10	General Purpose
Accumet Metall			F 100	4.00	40	ODD
13-620-115°	Platinum Half Cell, Glass Body		-5–100	140	10	ORP measurements, redox titrations
13-620-122°	Silver Billet Half Cell, Glass Body		-5–100	140	10	Silver and halide titrations
13-620-149 ^{P/P}	Dual Platinum Plate Combination		10-80	114	13	Chlorine titrations with CI Titrimeter®
13-620-123 ^{P/P}	Dual Platinum Pin Combination		-5-100	140	13	KF, dead-stop titra-
	GUITIDITIALIUII		-5-100	140	10	tions, sulfur analyses Environmental ORP

^A With U.S. Standard connector (and Pin connector on combinations). ^a With BNC connector. ^a Waterproof; BNC and Mini-Phone ATC connectors. ^a With DIN connector. ^a With Pin connector. ^a With Pin connectors. ^a With Twist-Lock waterproof connector.

Catalog No.	Description	Range	Interferences	pH/Temp./	L x Dia.	Typical Applications
		M/(ppm)		Range	(mm)	
cumet® Poly	mer Membrane Ion Sel		•			
3-620-537^	Calcium Combination,	10 ⁻⁶ to 10 ⁻²	Mg++, Zn++, Ba++,	3 to 10pH	102 x 13	Foods, beverages, soil
13-620-536 ⁸	Epoxy Body	0.04 to 400	K+, Na++, Ni++,	0° to 40°C		pharmaceuticals, explo-
			Cu ⁺⁺ , Fe ⁺⁺ , Sr ⁺⁺ ,			sives, fertilizers, plants
			H++, Hg++, Pb++			EDTA titration endpoin
13-620-535*	Nitrate Combination,		Cl ⁻ , NO ₂ ⁻ , Br ⁻ , CN ⁻ ,	2.5 to 11pH	102 x 13	Pollution testing, foods
13-620-534 ⁸	Epoxy Body	0.5 to 62,000	CIO ₃ -, I-, CIO ₄ -	0° to 40°C		pharmaceuticals, explo-
						sives, fertilizers, plants
13-620-533 ^A	Datassium Cambination	10-6 to 1	Cot NIII to The Life	0.E to 11 p.l.l	100 v 10	meats, pickling baths
13-620-532 ⁸	Potassium Combination, Epoxy Body	10 ⁻⁶ to 1 0.04 to 39,000	Cs+, NH ₄ +, TI+, H+, Ag+, +Tris+, Li+,	2.5 to 11pH 0° to 40°C	102 x 13	Body fluids, soils, sew- age, fertilizers, foods
13-020-332	choxy pony	0.04 (0.39,000	Ay, +IIIS, LI, Na+	0 10 40 0		beverages
A	l-State Ion Selective Elec		INA			Deverages
Accumer <i>30110</i> 13-620-520 ^A	Bromide Half Cell,	5 x 10 ⁻⁶ to 1	I⁻, CN⁻, S⁼	0 to 14pH	102 x 13	Biological fluids, soil
13-620-521 ⁸	Epoxy Body	0.4 to 79,000	1, 011, 3	0° to 14pm	102 X 13	plants, water, effluents
13-620-524 ^a	Bromide Combination,	5 x 10 ⁻⁶ to 1	S ⁼ , I ⁻ , CN ⁻ , high	2 to 14pH	108 x 13	foods. Method ASTN
13-620-525°	Glass Body		levels of Cl-, NH ₃	0°to 80°C	100 X 13	approved
13-620-518	Chloride Half Cell,	5 x 10 ⁻⁵ to 1	Br-, I-, CN-, S=,	0 to 14pH	102 x 13	Water/wastewater, soil
13-620-519 ⁸	Epoxy Body	1.8 to 35,500	OH-	0° to 100°C	102 X 10	dairy, tomato/vegetable
13-620-526^	Chloride Combination,	5 x 10 ⁻⁵ to 1	S=, I-, CN-, OH-,	2 to 12pH	108 x 13	products, meats. Meth-
13-620-527 ⁸	Glass Body	1.8 to 35,500	Br	0° to 80°C		od ASTM/AOAC approved
13-620-540^	Cupric Half Cell,	10 ⁻⁸ to 10 ⁻¹	Ag⁺, Hg ⁺⁺ , Cl⁻,	2 to 12pH	102 x 13	Plating baths
13-620-541 ⁸	Epoxy Body	6.4x 10 ⁻⁴ to 6350	Br-, Fe+,Cd++	0° to 80°C		natural water
13-620-546^	Cupric Combination,	10 ⁻⁸ to 10 ⁻¹	Ag+, Hg++, Cl-,	2 to 12pH	108 x 13	silicor
13-620-547 ⁸	Glass Body	6.4x 10 ⁻⁴ to 6350	Br⁻, Fe⁺,Cd++	0° to 80°C		
13-620-539^	Cyanide Combination,	5 x 10 ⁻⁶ to 10 ⁻²	Cl ⁻ , Br ⁻ , l ⁻ ,	11 to 13pH	102 x 13	Petrochemical, plating
13-620-538 ⁸	Epoxy Body	0.1 to 260	S= absent	0° to 80°C		water, wastes
13-620-522	Fluoride Half Cell,	Sat. to 10 ⁻⁶	OH-	5 to 9pH	102 x 13	Water/wastewater, stack
13-620-523 ⁸	Epoxy Body	Sat. to 0.02		0° to 80°C		gases, explosives, etch-
						ing baths, bioresearch
13-620-528 ^A	Fluoride Combination,	Sat. to 10 ⁻⁶	OH-	5 to 8pH	108 x 13	beverages, detergents Method ASTM/EPA
13-620-529°	Glass Body	Sat. to 0.02	UH	0° to 80°C	100 X 13	approved
13-620-542	Lead Half Cell,	10 ⁻⁶ to 10 ⁻¹	Ag+, Hg++, Cu++,	3 to 8pH	102 x 13	Organic compounds
13-620-543 ⁸	Epoxy Body	0.2 to 20,700	Cd++, Fe++	0° to 80°C	102 X 10	biological samples
13-620-548	Lead Combination,	10 ⁻⁶ to 10 ⁻¹	Ag+, Hg++, Cu++,	3 to 8pH	108 x 13	water/wastewater
13-620-549 ⁸	Glass Body	0.2 to 20,700	Cd++, Fe++	0° to 80°C		,
13-620-544	Silver/Sulfide Half Cell,	10 ⁻⁷ to 1.0 (Ag ⁺ /S ⁼)		2 to 12pH	102 x 13	Water, pulping
13-620-545 ⁸	Epoxy Body	0.01 to 107,900 (Ag		0° to 80°C		liquors
13-620-550^	Silver/Sulfide Combi-	0.003 to 32,100 (S	=)	2 to 12pH	108 x 13	
13-620-551 ⁸	nation, Glass Body			0° to 80°C		
Accumet Gas-	Sensing Ion Selective E	lectrodes •				
13-620-504	Ammonia Combination,	5 x 10 ⁻⁷ to 10 ⁻¹	Volatile amines,	13pH	108 x 18	Sewage effluent, boiler
13-620-505 ⁸	Epoxy Body	0.009 to 1700	metal cations	0° to 50°C		water, industrial waste
			that complex			stack gases, food, ferti-
			ammonia			lizers. Method ASTM,
						EPA approved
13-620-506^	Carbon Dioxide Combi-	10 ⁻⁵ to 3 x 10 ⁻²	Volatile organ-	4.5pH	108 x 18	Measures carbon di-
13-620-507 ⁸	nation, Epoxy Body	0.440 to 1320	ic acids	0° to 50°C		oxide, carbonate, bicar-
						bonate in beverages
1		. =!				wines, ground/sea water
	s-Membrane Ion Selecti				100 :-	
13-620-500 ^A	Sodium Half Cell,	10 ⁻⁶ to 10 ⁰	Ag+, Li+, K+,	0 to 14pH	102 x 13	Meats, fish, dairy pro-
13-620-501 ⁸	Glass Body	0.023 to 23,000	NH_4^+	0° to 80°C		ducts, fruit juices
13-620-502 ^A	Codium Combination	10 ⁻⁶ to 10⁰	Λα÷ Li÷ V÷	0 to 145U	1111 v 10	brewing water, ground
13-620-503 ⁸	Sodium Combination, Epoxy Body	0.023 to 23,000	Ag+, Li+, K+, NH ₄ +	0 to 14pH 0° to 60°C	114 x 13	water, sea water, soils bodily fluids

Catalog No.	Description	pH Range	Temp. Range (°C)	Length (mm)	Diameter (mm)	Typical Applications
Accumet® Ann	ular Junction Reference Ha	If Cells For I	ISEs •			
13-620-46°	Single Annular Junction, Epoxy Body, Ag/AgCl	0–14	0–100	108	13	Calcium, cyanide, divalent cation, fluoride, sodium, redox, pH
13-620-47°	Double Junction (Annular/ Ceramic), Epoxy Body, Ag/AgCl inner, outer cham- ber empty for use with sample-compatible electrolyte	0–14	0–100	108	13	Bromide, chloride, cop- per, iodide, lead, nitrate, silver/sulfide; redox and pH applications requiring compatible electrolyte
Cell Constant	2-Cell Mo	dels Cat. No	D.	4-Ce	II Models Cat. I	No.
	Glass Body	Ep	oxy Body	Glass Body		Epoxy Body

13-620-161P/P

13-620-160P/P

13-620-162P/P

13-620-163^{D/M}

13-620-164D/M

13-620-165^{D/M}

13-620-166^{D/M}

13-620-157^{P/P} Mith DIN and Mini-ATC connector. PWith Pin connector. PPWith Dual Pin connectors.

13-620-156P/P

13-620-155P/P

0.1cm⁻¹ 1.0cm⁻¹

10.0cm⁻¹

Accumet® Immersion Type Conductivity Cells For Accumet AB30, AR20, and AR50 Meters

pН	Color	Ingredients	Size	Catalog No.
pH Buff	ers			
4.00	Red	Potassium Biphthalate	500mL	SB101-500
7.00	Yellow	Potassium Phosphate Monobasic & Sodium Hydroxide	500mL	SB107-500
10.00	Blue	Potassium Carbonate, Potassium Borate & Potassium Hydroxide	500mL	SB115-500
4, 7, 10		Fisher Buffer-Pac: 500mL ea. of color-coded pH 4, 7, and 10 buffers	3x500mL	SB105
4.00	Red	Individual Tear-open pH Packets	20/box	SB4
7.00	Yellow	Individual Tear-open pH Packets	20/box	SB7
10.00	Blue	Individual Tear-open pH Packets	20/box	SB10
Description			Size	Catalog No.
		and Storage		
Electrode Rinse Solution in Individual			20/box	SB15
Tear-open Packets; color-coded Gray			41	0540.4
Electrode Storage Solution Electrode Storage Bottle			1L	SE40-1 13-620-499
Electrone	Sluraye c	oulle		13-020-499
Description			Size	Catalog No.
Electroc	le Filling	Solutions		
Saturate			500mL	SP138-500
		gle-junction electrodes;		
		of all Ag/AgCl double-junction,		
		uTupH® and AccuTupH+® electrodes.	F00I	00405 500
		d with AgCl.	500mL	SP135-500
	allall cin	gle-junction electrodes; for inner		

ISE	Solution	Size(mL)	Catalog No.
Standards and	Solutions for Accumet ISEs		
Ammonia	0.1M NH₄Cl	500	13-620-800
	1000ppm as NH₃	500	13-620-801
	pH/ISA; 10M NaOH	500	13-620-802
	Filling solution	500	13-620-803
Bromide	0.1M NaBr	500	13-620-821
	1000ppm Br	500	13-620-822
	ISA; 5M NaNO₃	500	13-620-823
Calcium	0.1M CaCl ₂	475	13-620-811
	100ppm Ca ⁺⁺	475	13-620-862
	ISA; 4M KCI	475	13-620-851
Carbon Dioxide	0.1M NaHCO₃	500	13-620-804
	ISA	500	13-620-805
	Filling solution	500	13-620-806
Chloride	0.1M NaCl	500	13-620-818
	1000ppm Cl ⁻	500	13-620-819
	ISA; 5M NaNO ₃	500	13-620-820
Cyanide	ISA; 10M NaOH	500	13-620-802
Fluoride	0.1M NaF	500	13-620-824
	1000ppm F ⁻	500	13-620-825
	TISAB	500	13-620-831
	TISAB II	500	13-620-835
Nitrate	0.1M NaNO ₃	475	13-620-888
	1000ppm N	475	13-620-910
	100ppm N	475	13-620-924
	ISA; 2M (NH ₄) ₂ SO ₄	475	13-620-850
Potassium	0.1M KCI	475	13-620-917
	ISA; 5M NaCI	475	13-620-927
Sodium	10% NaCl	500	13-620-826
	100ppm as Na ⁺	500	13-620-827
	1000ppm as Na ⁺	500	13-620-828
	ISA	500	13-620-832



Accu • pHast and AccuTupH electrodes.

Fisher Scientific Pittsburgh, PA (412) 490-8300

Internet address: www.fishersci.com

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